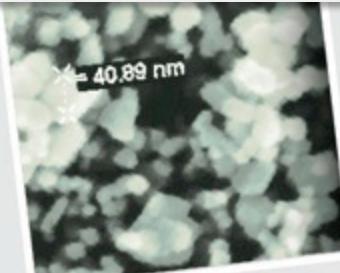


NCI
Southwest

Occurrence, Removal and Regulation of Nanomaterials at Publicly Owned Sewage Treatment Works

Thursday, May 26, 2016

[CLICK HERE FOR THE WEBINAR RECORDING](#)





Occurrence, Removal and Regulation of Nanomaterials at Publicly Owned Sewage Treatment Works

Begin Recording

Paul Westerhoff, PhD, PE, BCEE

School of Sustainable Engineering and
The Build Environment

Ira A. Fulton Schools of Engineering
Arizona State University (Tempe, AZ)

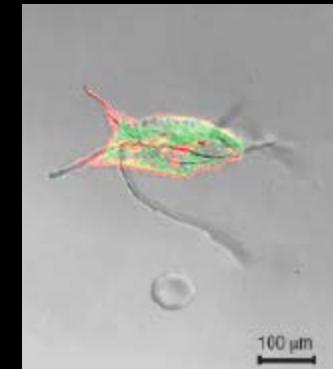
Benefits & Risks of Nanotechnology

**GOOD
nano**



Quantum size effects result in unique mechanical, electronic, photonic, and magnetic properties

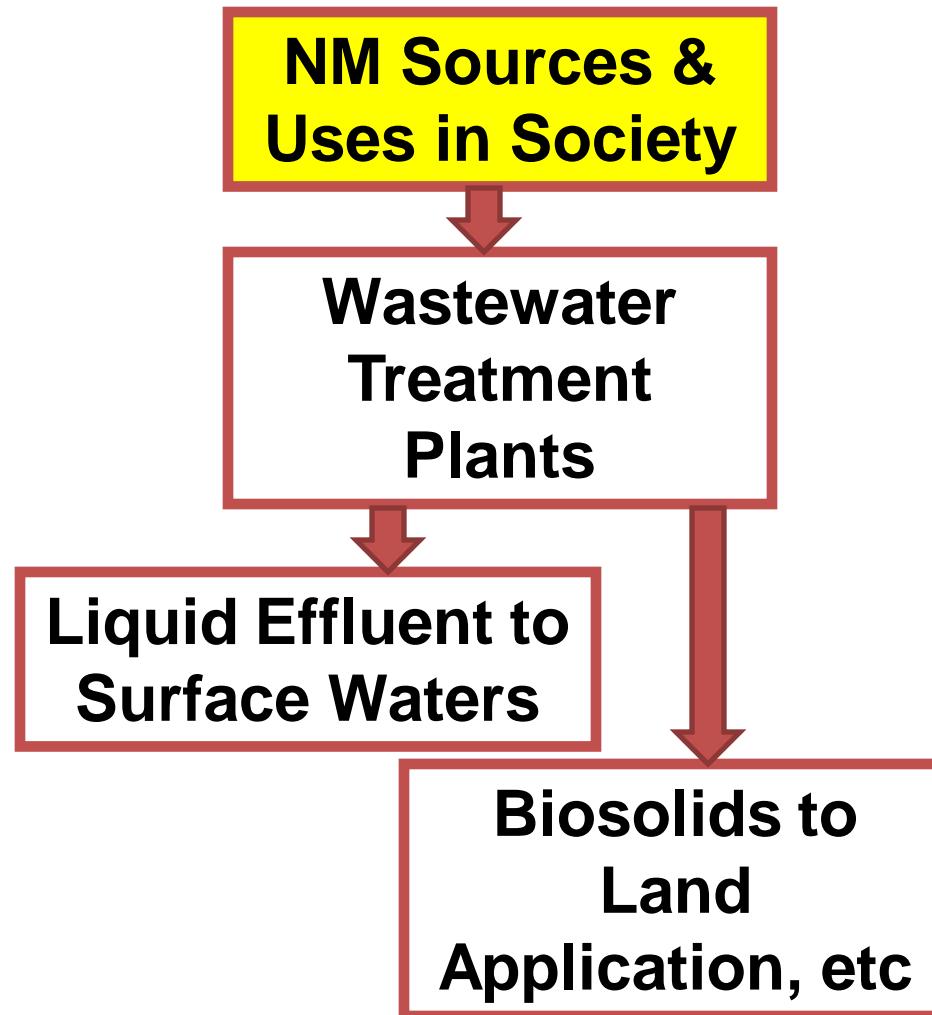
**BAD
nano**



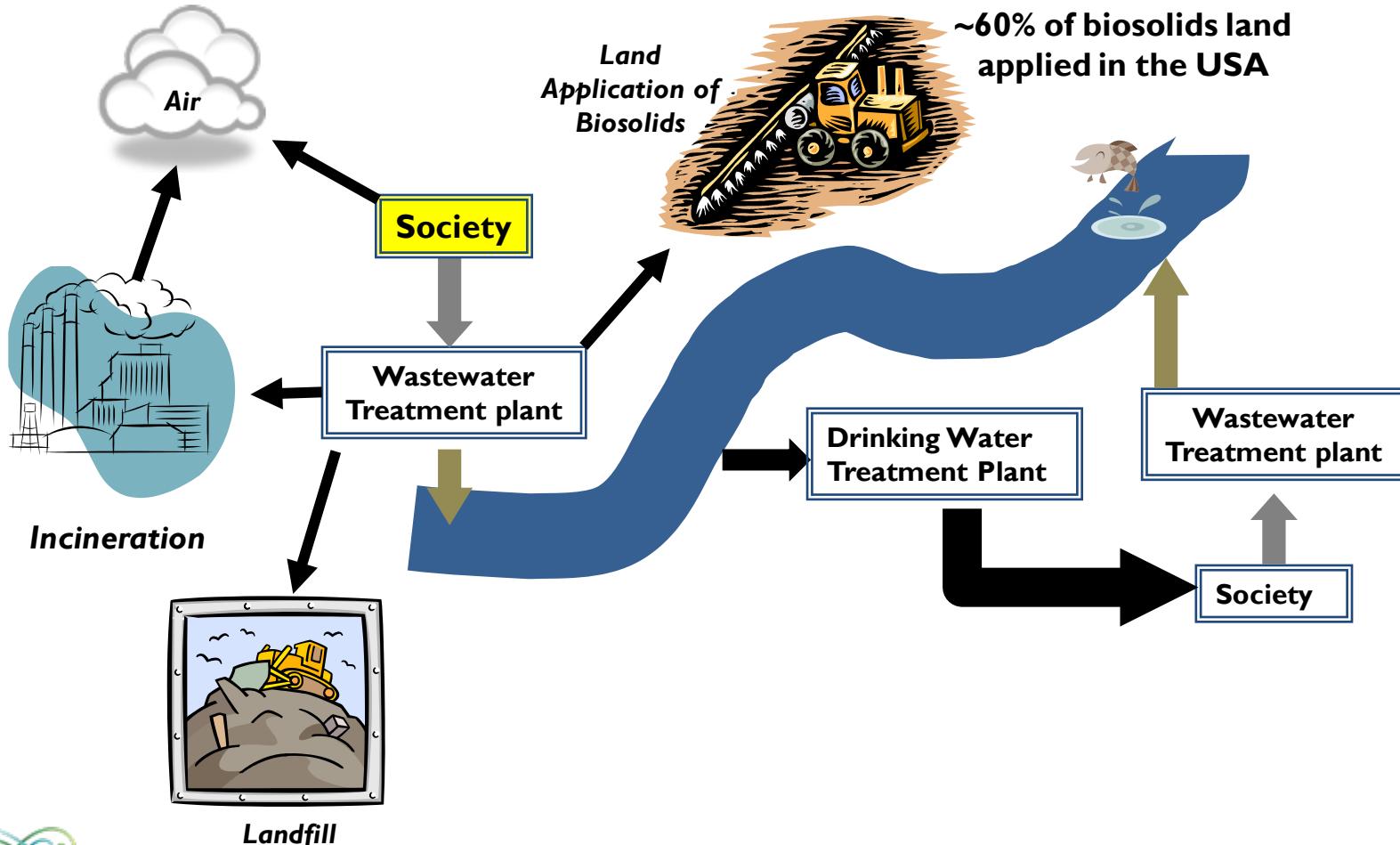
Nanoparticles operationally defined as < 100 nm in at least one-dimension



Presentation Outline



Society is a source of incidental & engineered Nanomaterials

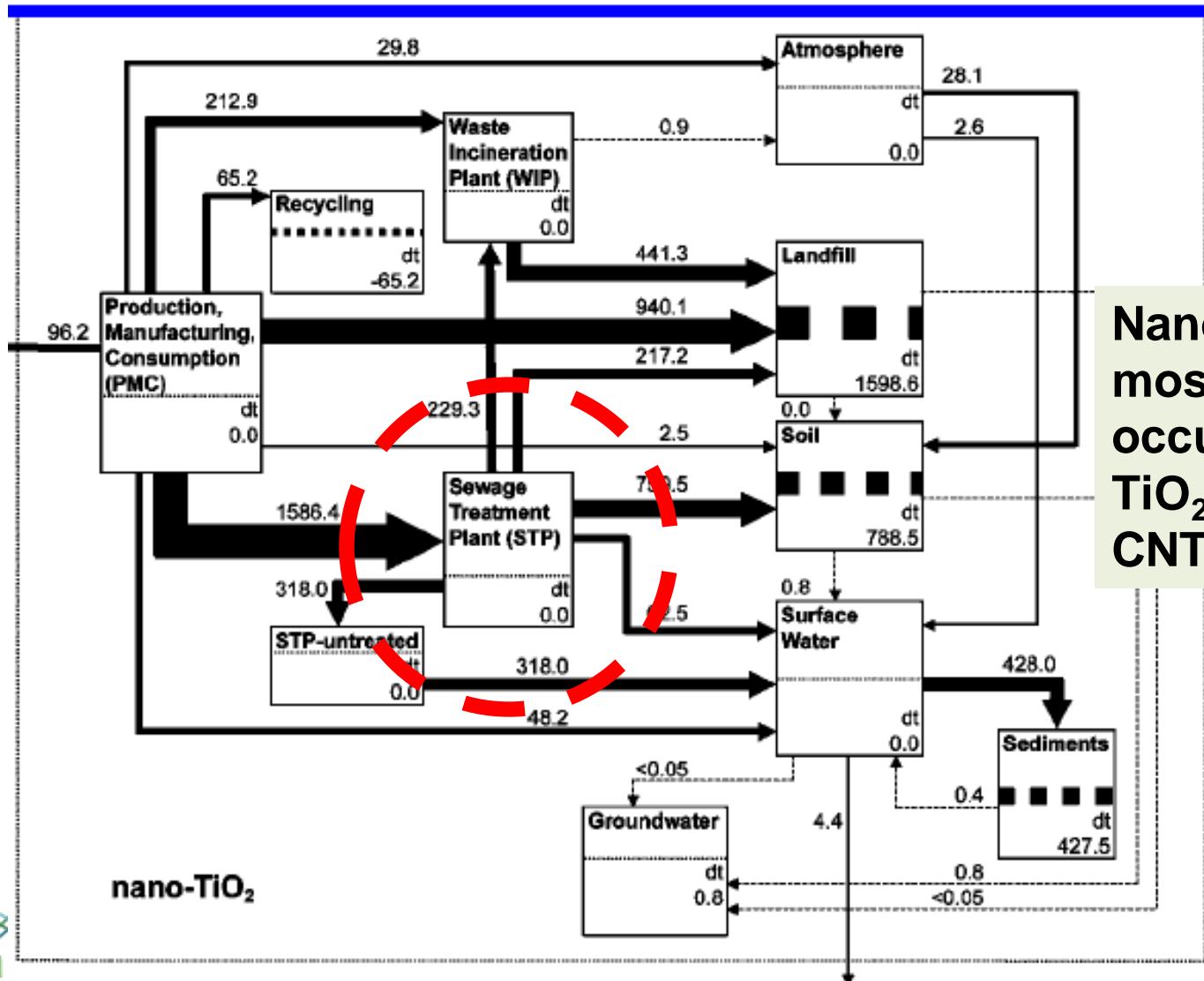




Projections indicate high NM loadings to Sewers & WWTPs

ASU

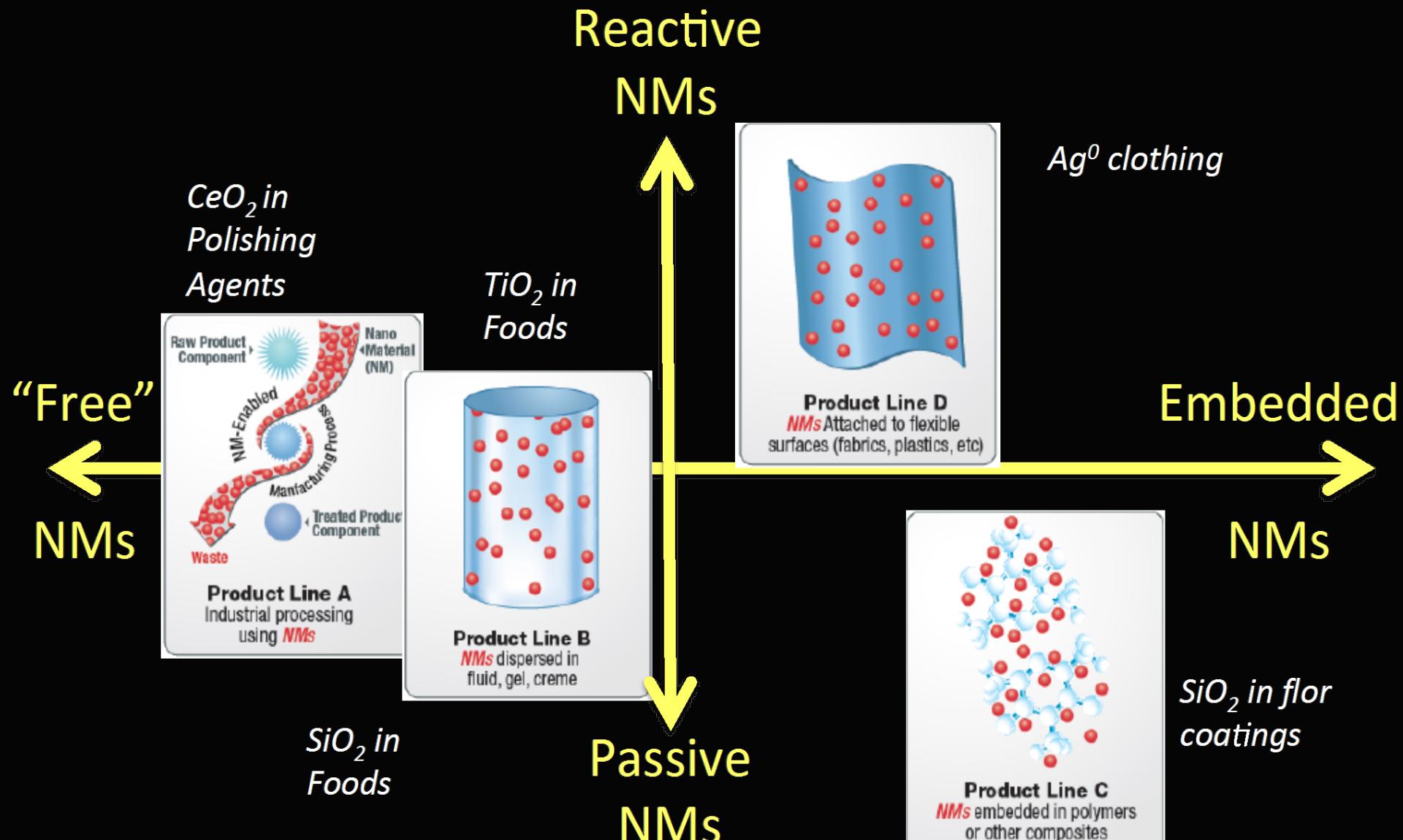
Environ. Sci. Technol. 2009 43, 9216–9222



Nanomaterials most likely to occur:

$\text{TiO}_2 > \text{ZnO} > \text{Ag} > \text{CNT} > \text{fullerene}$

LCnano Product Lines include NM Functionality & Application





Product Line A

Polishing Agents

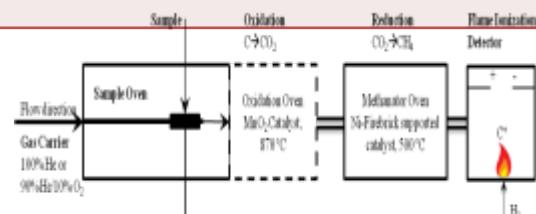
- Industrial processes for electronics, optics etc use polishing agents in fluids
- >>\$1B industry
- SiO_2 , Al_2O_3 , CeO_2 are common polishing agents
- Polishing agents end up in sewers – not consumer products
- No direct regulation exists for their discharge to sewers





Common Direct NM Detection

Method	NM Type	Generalized Detection Limit Comments
Light scattering (UV/VIS)	any	> 1 mg/L in water 0.05 mg/L with HPLC of NM extract
ICP-MS (better than ICP-AES) TOF-MS (emerging)	metals	> 10 ppt in water (total metal conc) Ability to get size, # and concentration dosimetry (sp-, FFF-, centrifugal – ICP-MS)
LC-MS	C60	~ 1 ppt
Thermal combustion or Microwave thermal analysis	MWCNT SWCNT	~ 1 ppb
Isotopes	¹⁴ C Metal isotopes	< 1 ppb by Scintillation counting Isotopic ratios



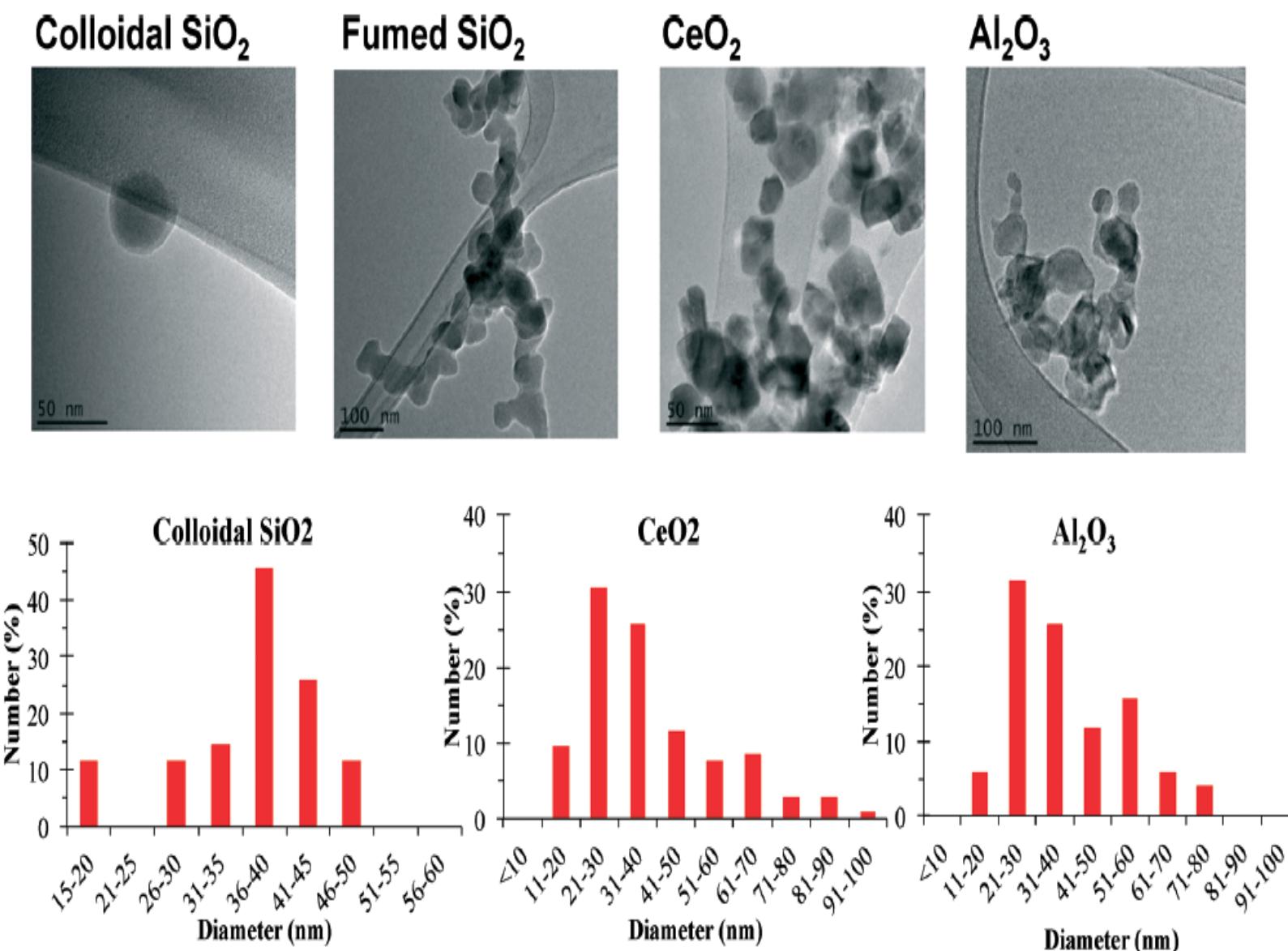
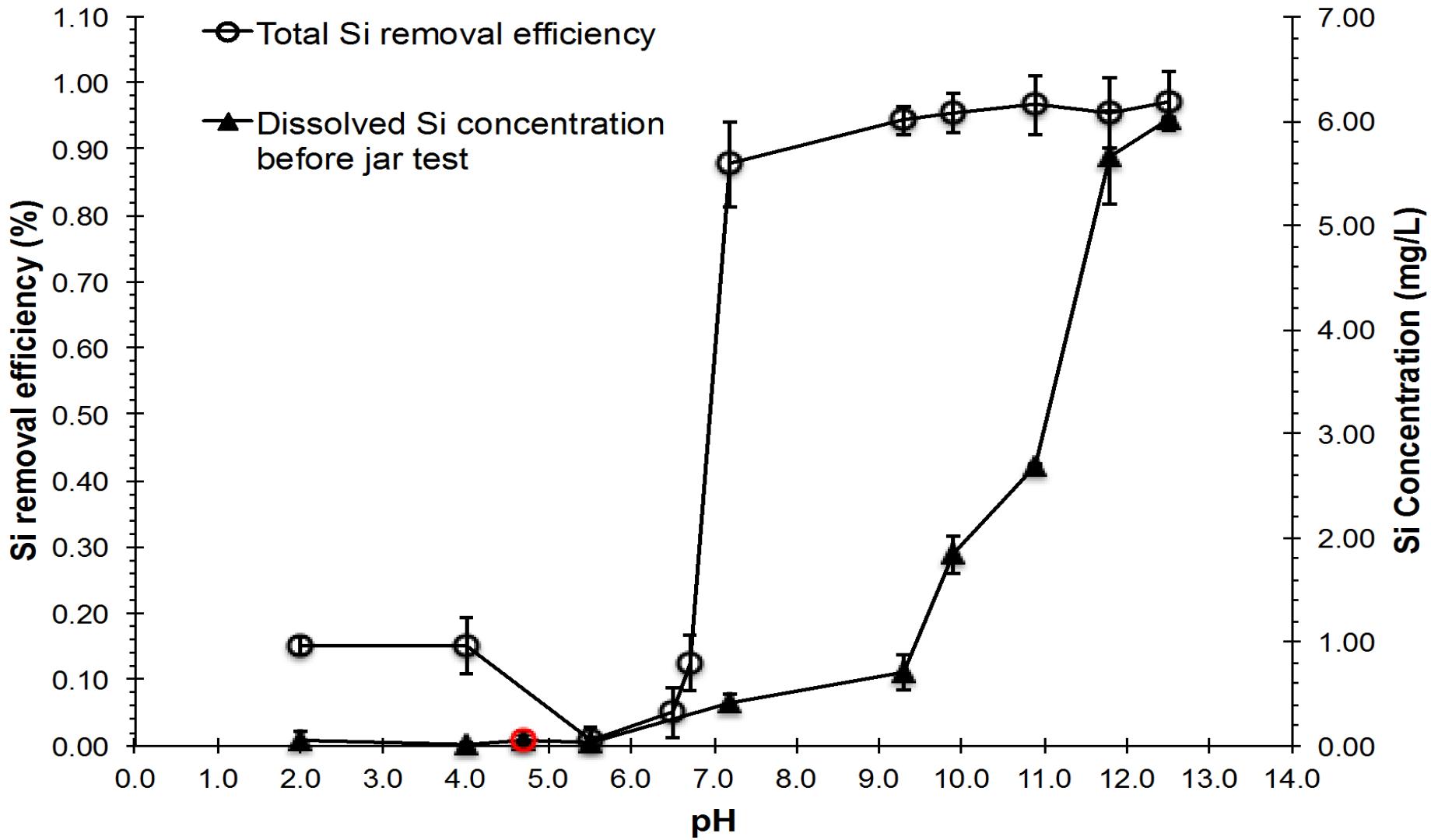
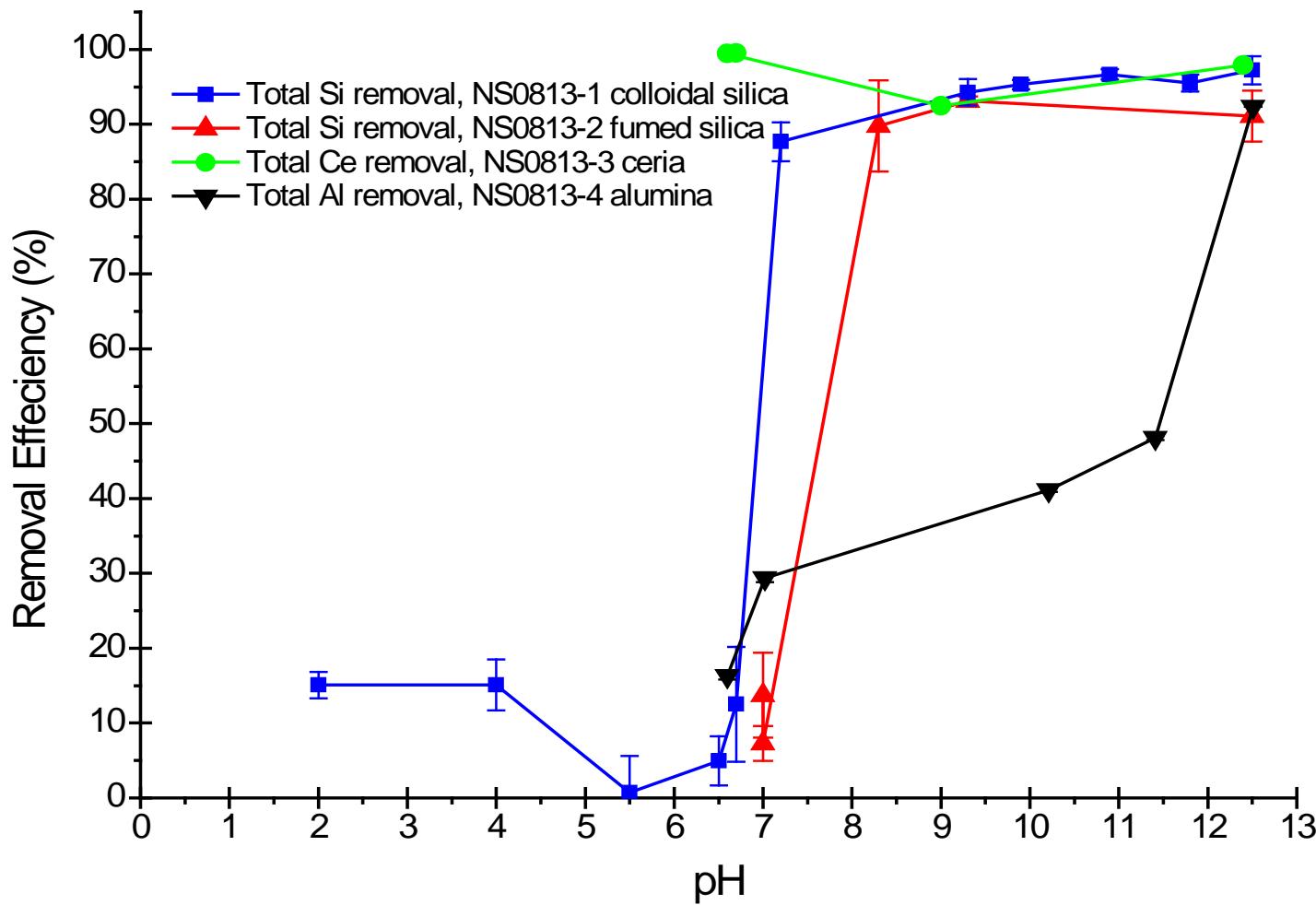


Fig. 6 TEM images and TEM based particle size distributions for CMP nanoparticles. The size distribution histogram for colloidal silica, ceria, or alumina is obtained by sizing >50 particles under the corresponding TEM images. Fumed silica particles were not sized because of their coalesced state.

On-site Chemical Lime Softening Treatment of CMP wastestreams





Removal efficiencies of Si, Ce and Al for 4 CMP slurries under different pH conditions. Ca dosage was 2mM in all cases. Started concentrations: $C_{0\text{Si,slurry}1}=24 \text{ mg/L}$, $C_{0\text{Si,slurry}2}=20 \text{ mg/L}$, $C_{0\text{Ce,slurry}3}=19.5 \text{ mg/L}$, $C_{0\text{Al,slurry}4}=31 \text{ mg/L}$. pH values are obtained after jar test when reactions are considered in equilibrium.

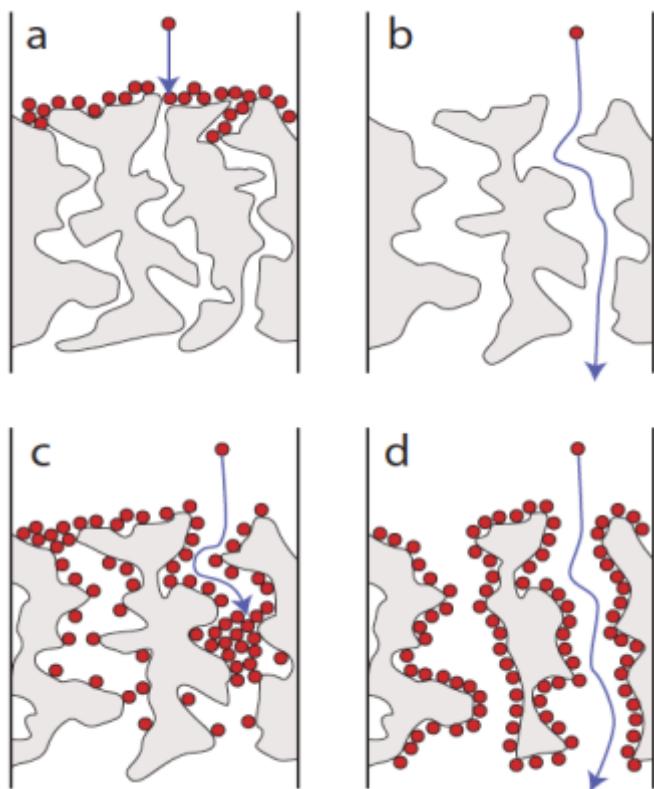
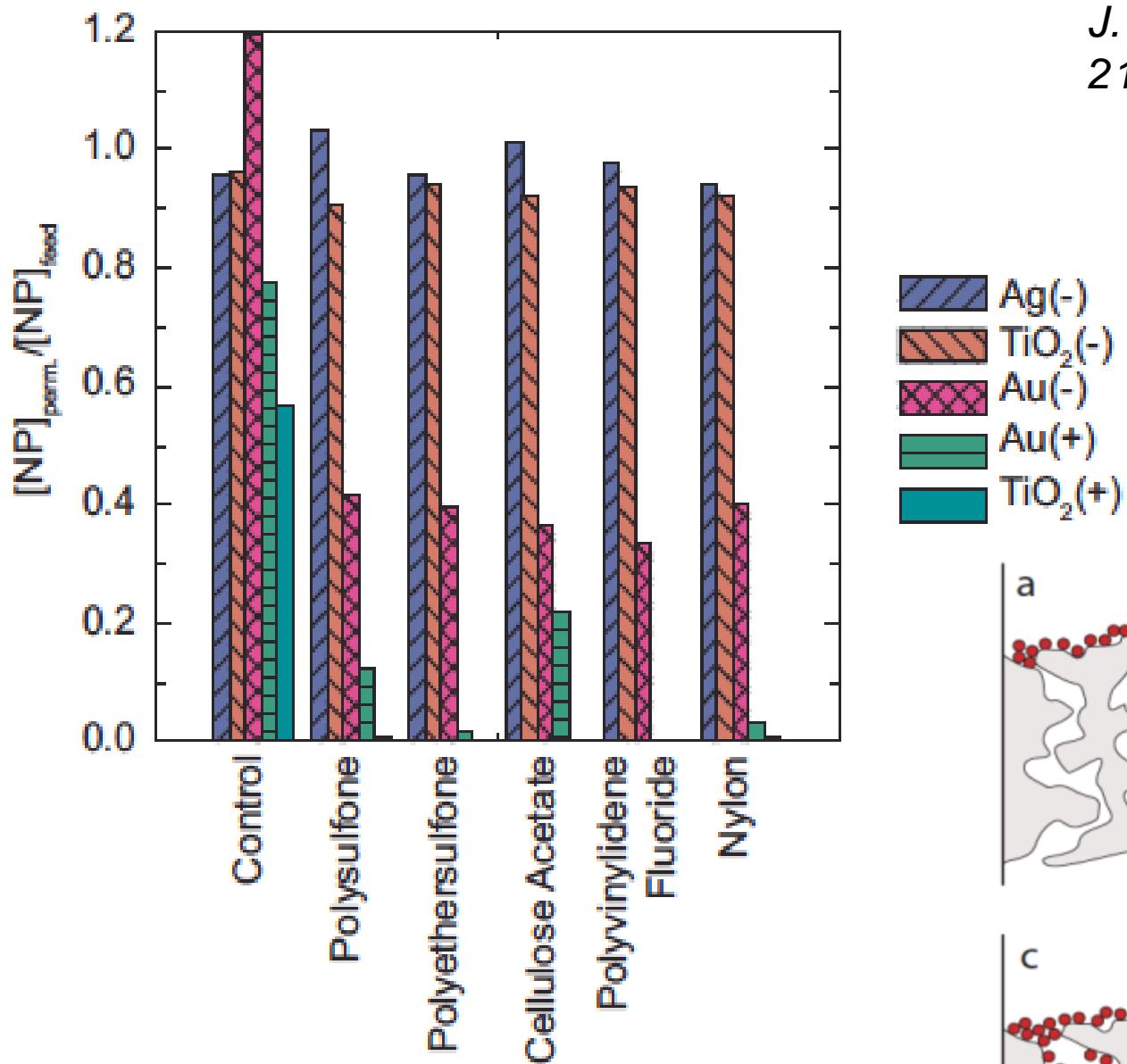


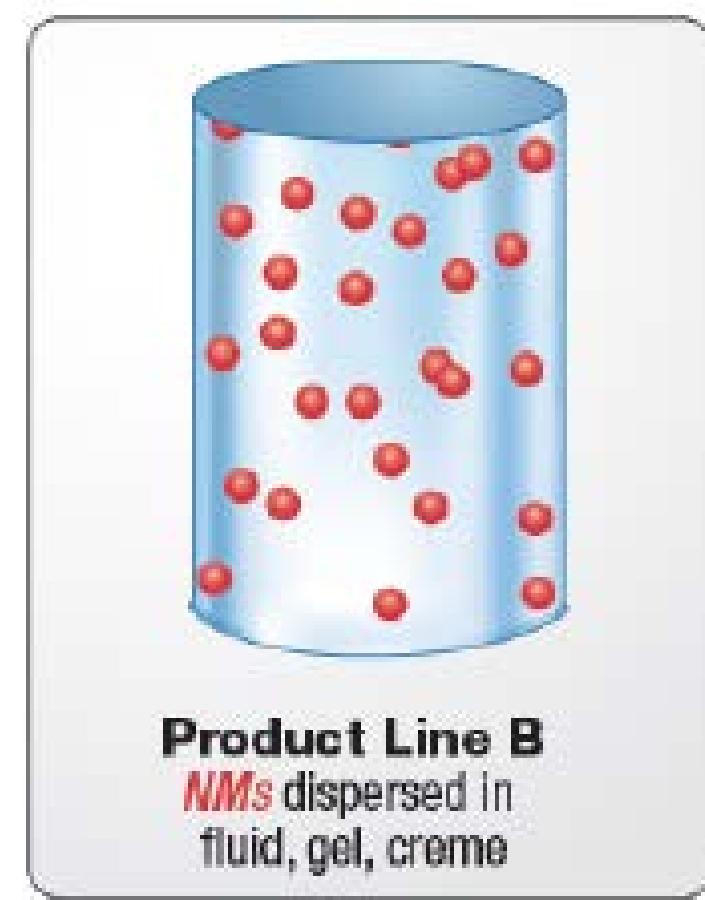
Fig. 3. Normalized permeate concentrations of NPs passed ten syringe filters (0.22-μm pore size) made of five different polymeric trol samples were passed through the filtration device with no men



Product Line B

Nanomaterials in foods

- Nanomaterials added to food for a variety of reasons:
 - Texture
 - Anti-caking
 - Color
 - Oxygen barrier
 - Abrasives
 - Antimicrobial
- Little to no regulation exist
- Difficult to monitor

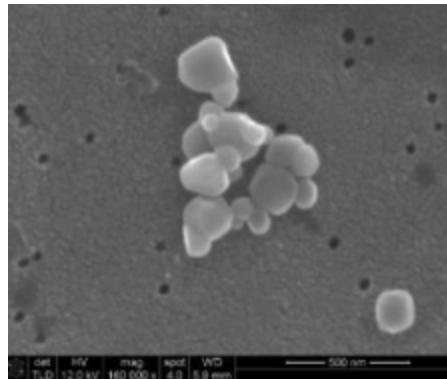
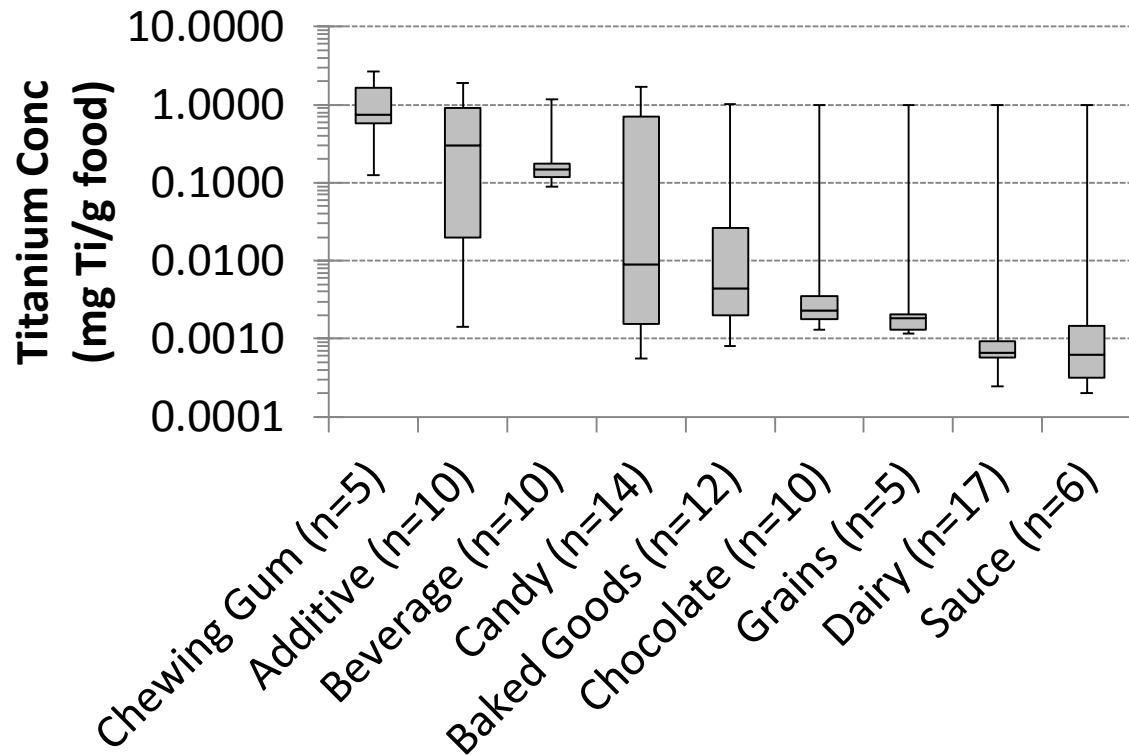


Nanomaterials Dispersed in Products

Nano-Ag in Toothpaste



Silver in Shampoo

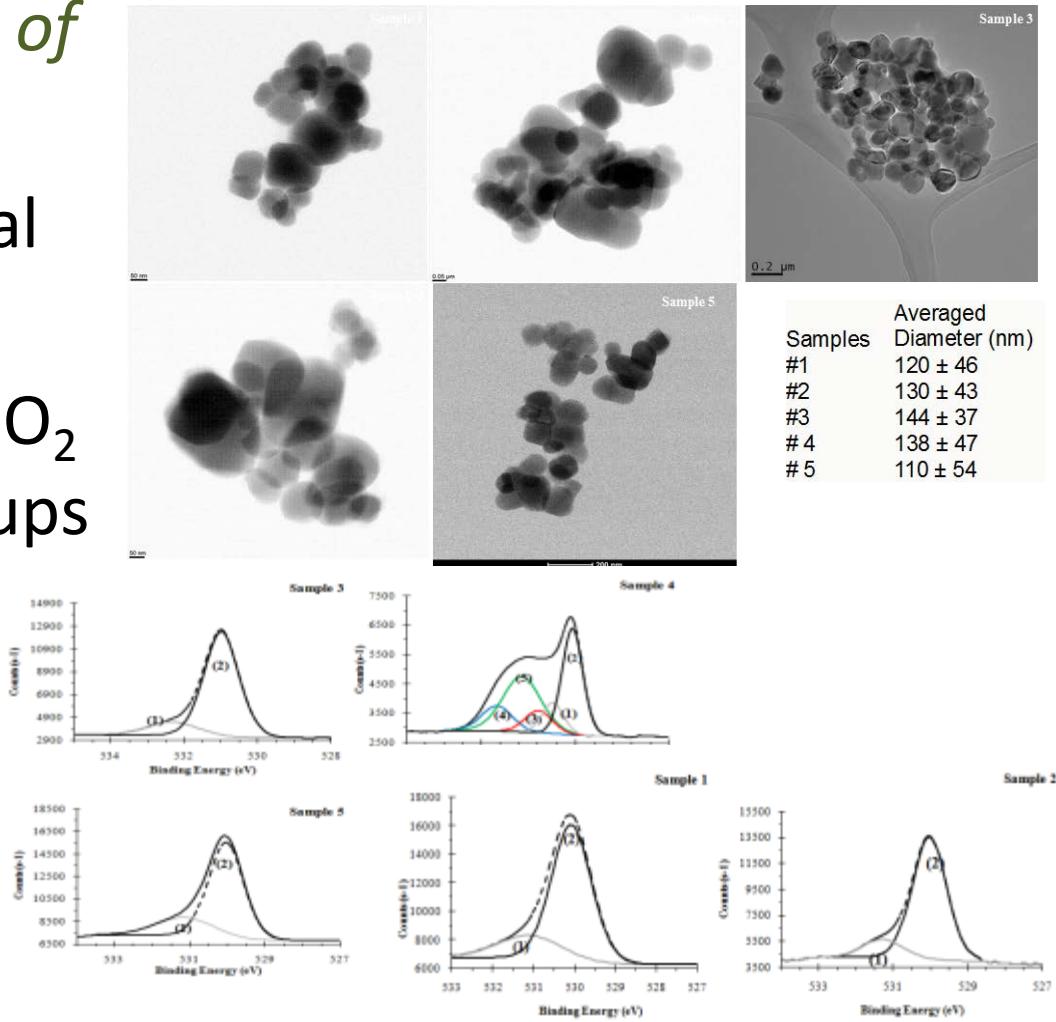
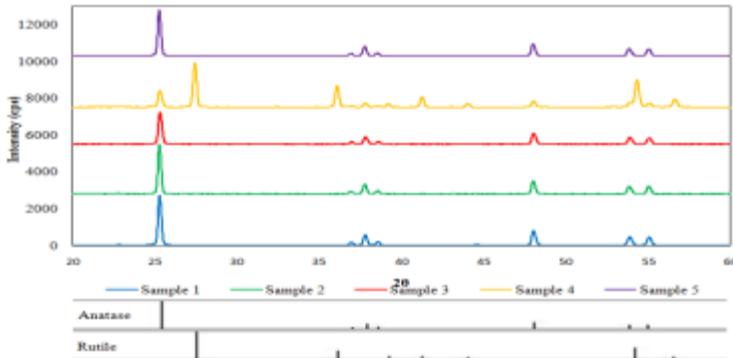


Weir et al.,
ES&T (2012)



Test thousands of products?

- *Go to suppliers instead of food grade products*
- Can assess fundamental properties of NPs
- Example: food grade TiO₂ contains P-surface groups & is primarily anatase





Beverages contain all types of nanoparticles

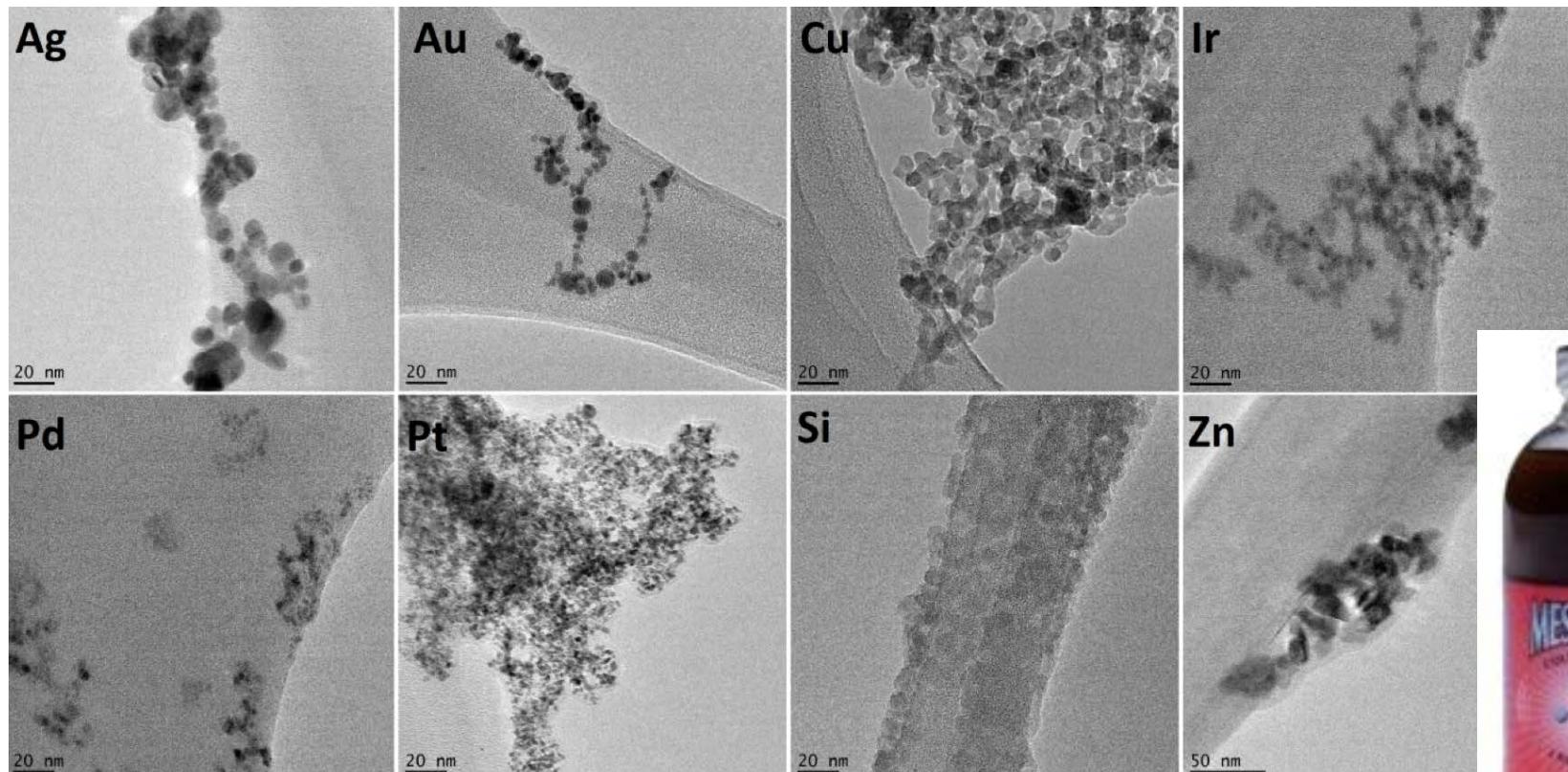


Figure 1. Representative TEM images of NMs found in supplement drinks.



Silica , Anti-caking agent in food



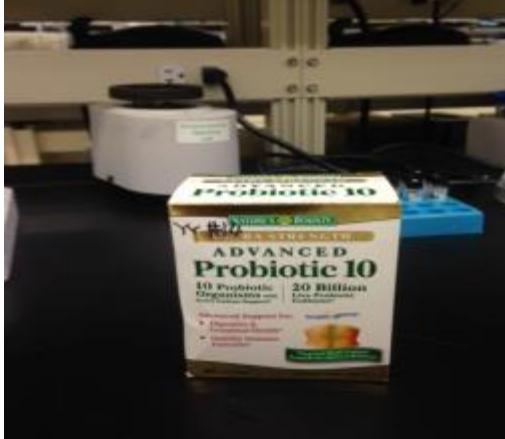
In the United States, the Food and Drug Administration allows up to 2% by weight of silica to foods as an anti-caking agent.

Nature Nanotechnology 9, 658–659 (2014) doi:10.1038/nnano.2014.196

http://www.reddit.com/r/FanTheories/comments/1caz5e/the_cookie_monsters_eating_disorder_and_failed/

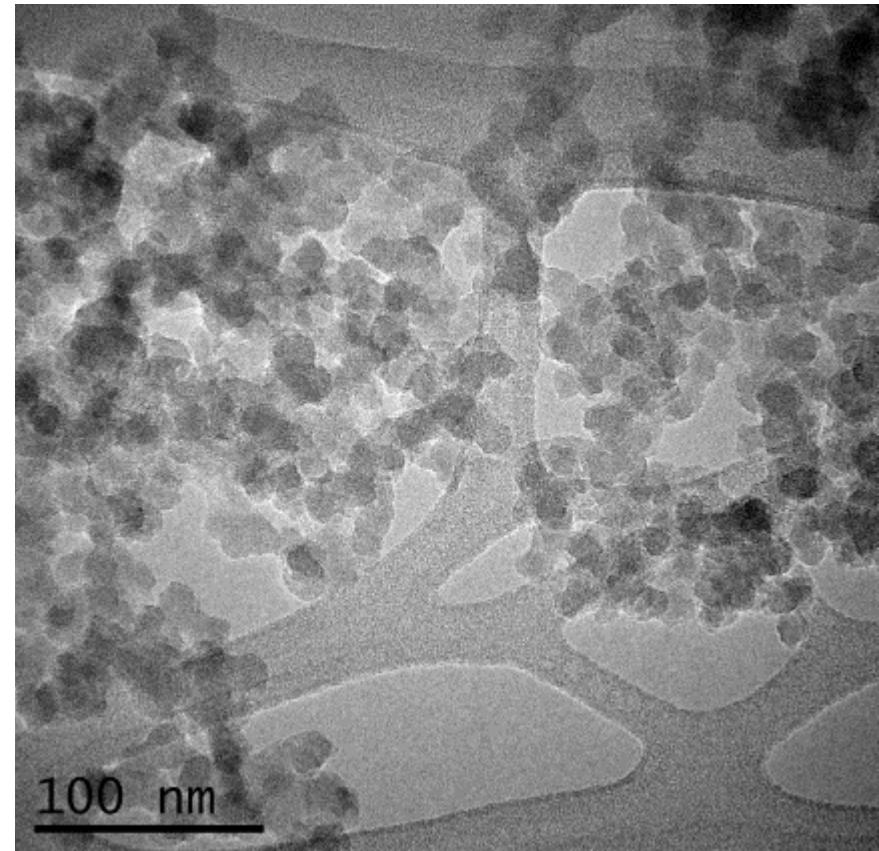
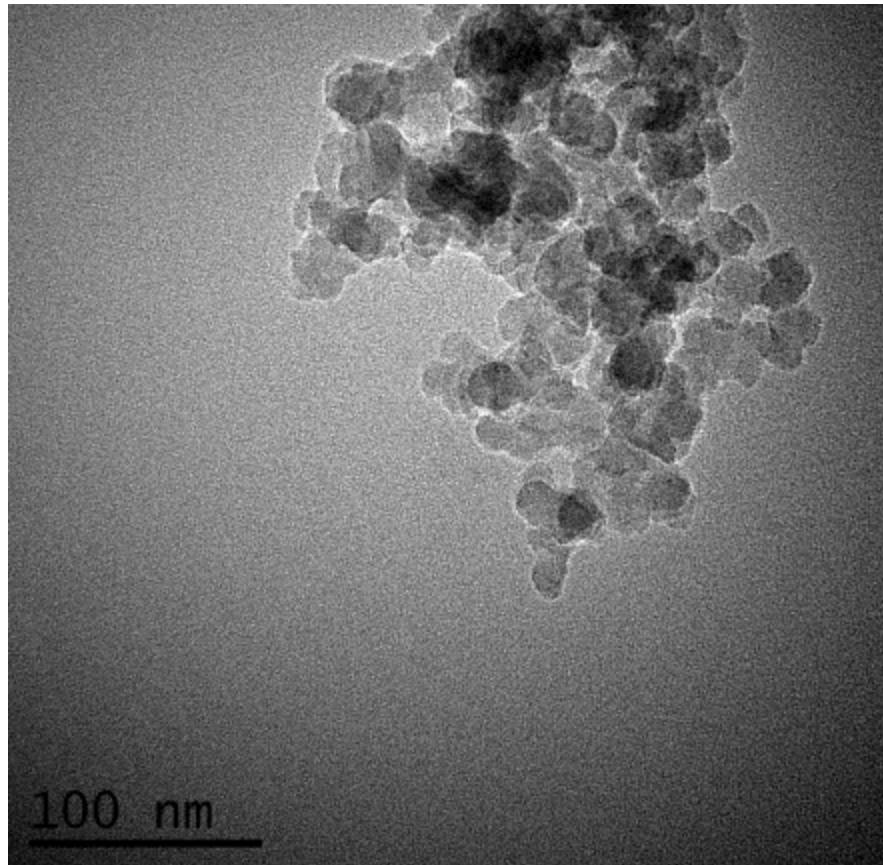


Foods Claim Containing Silica





Transmission Electron Microscopy (TEM) on Separated Silica from Foods



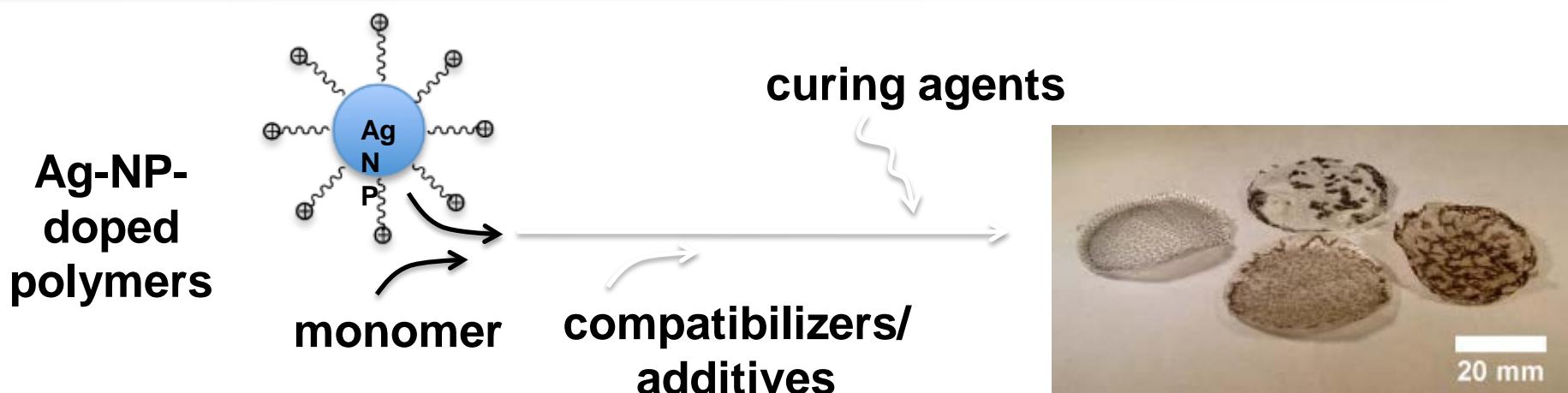
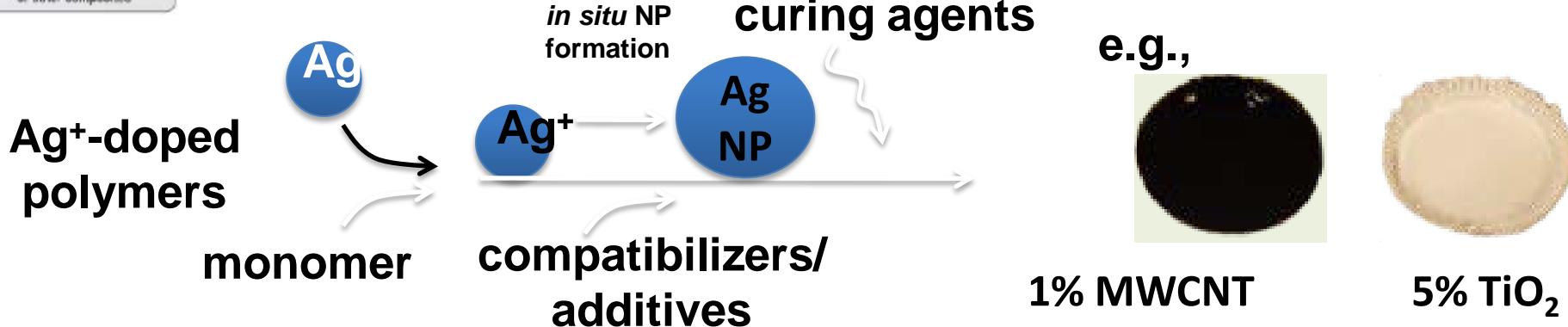
→ Nano silica present in foods



Product Line C
NMs embedded in polymers or other composites

Product Line C

Example: nano-Ag in plastics



Ag-impregnated Tupperware

amazon.com®

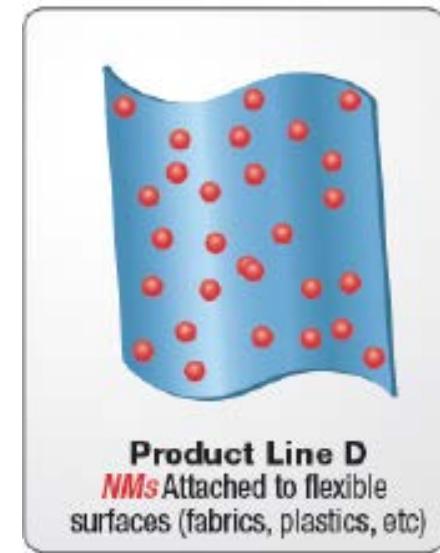


Sharper Image



Product Line D – Nano-enabled fabrics

- Why?
 - Anti-microbial (no stink)
 - Flame Retardant
 - Self cleaning
- Common nanomaterials:
 - Silver
 - Silica
 - CNT
- What happens during use (wearing & washing)?





AgNPs attached to textiles

Tethered-AgNP

Electrostatic-AgNP

Purchased Ag textiles

Ag⁰-coated

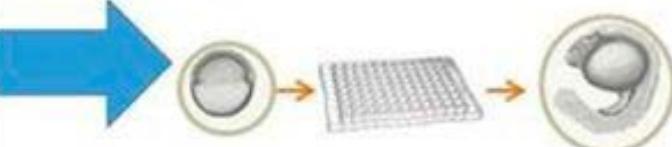
Ag salt-coated

Use phase: washing of Ag-textiles



Wash water exposure to zebrafish embryos

Fertilization 6 h 24 h (1 day)



Textiles after washing



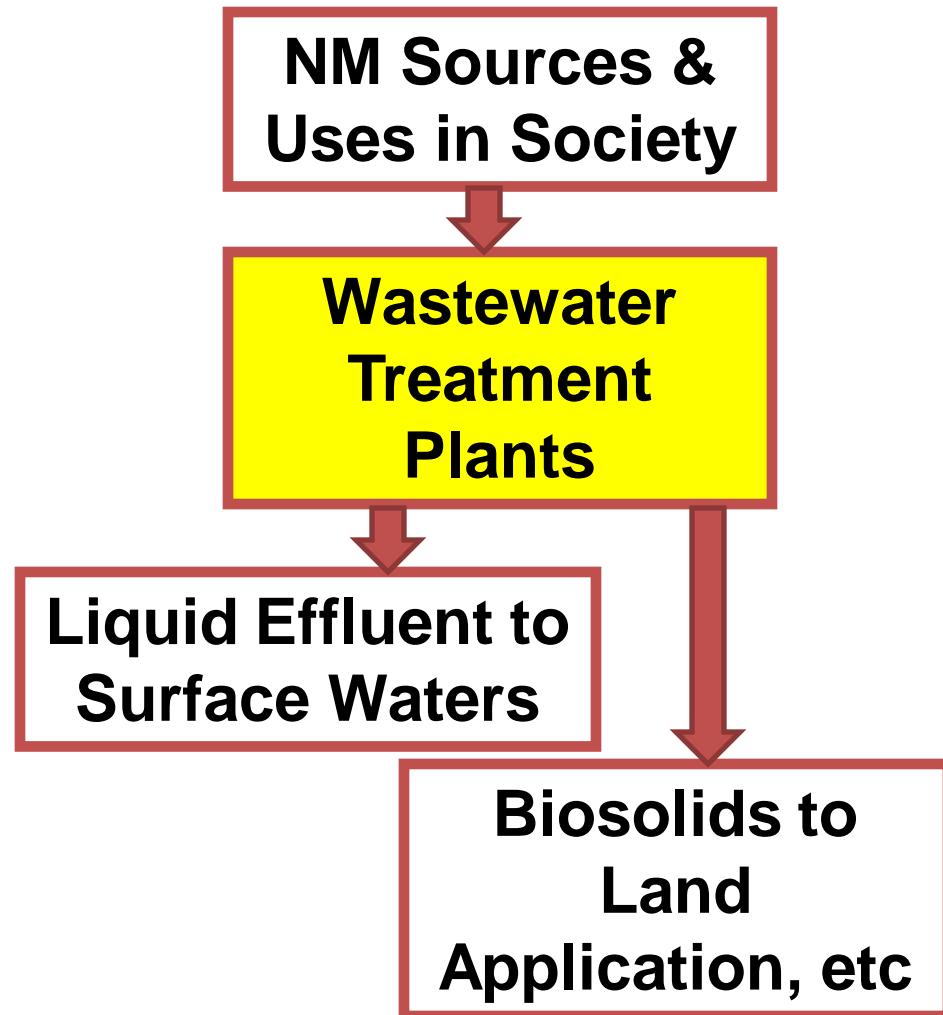
Disposal and end-of-life considerations

Ag-textiles antimicrobial efficacy

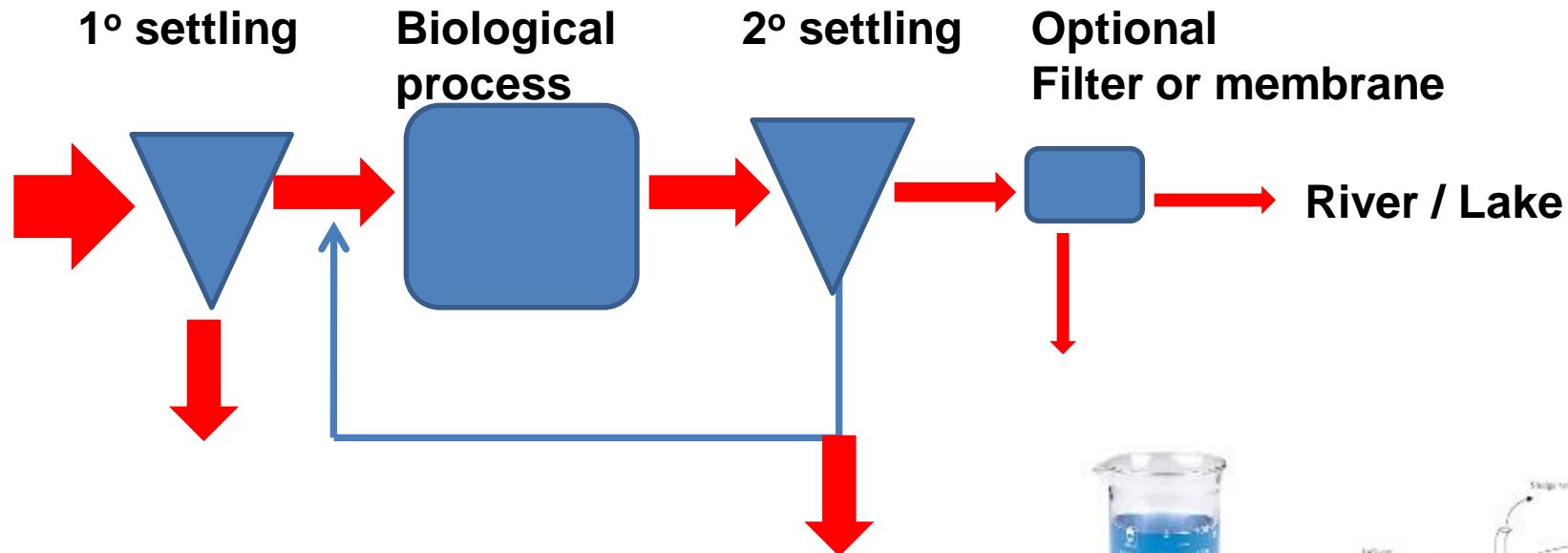


Part I - Summary

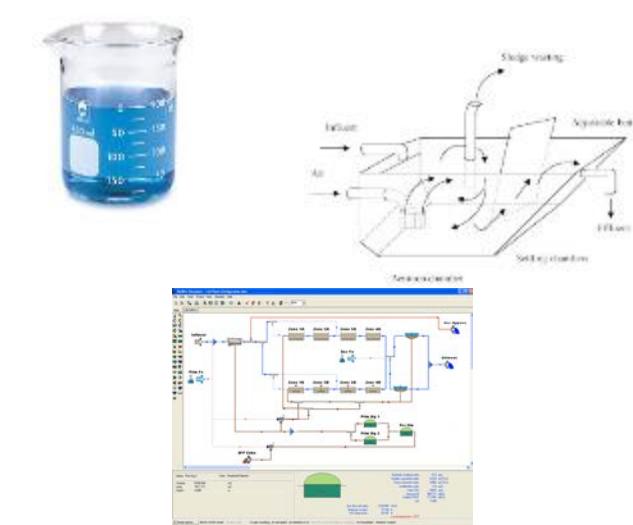
- NM s are released from commercial products into sewage wastewater
- The functionality of the product influences how NM s are used in different product lines



Nanomaterial removal in Wastewater Treatment Processes

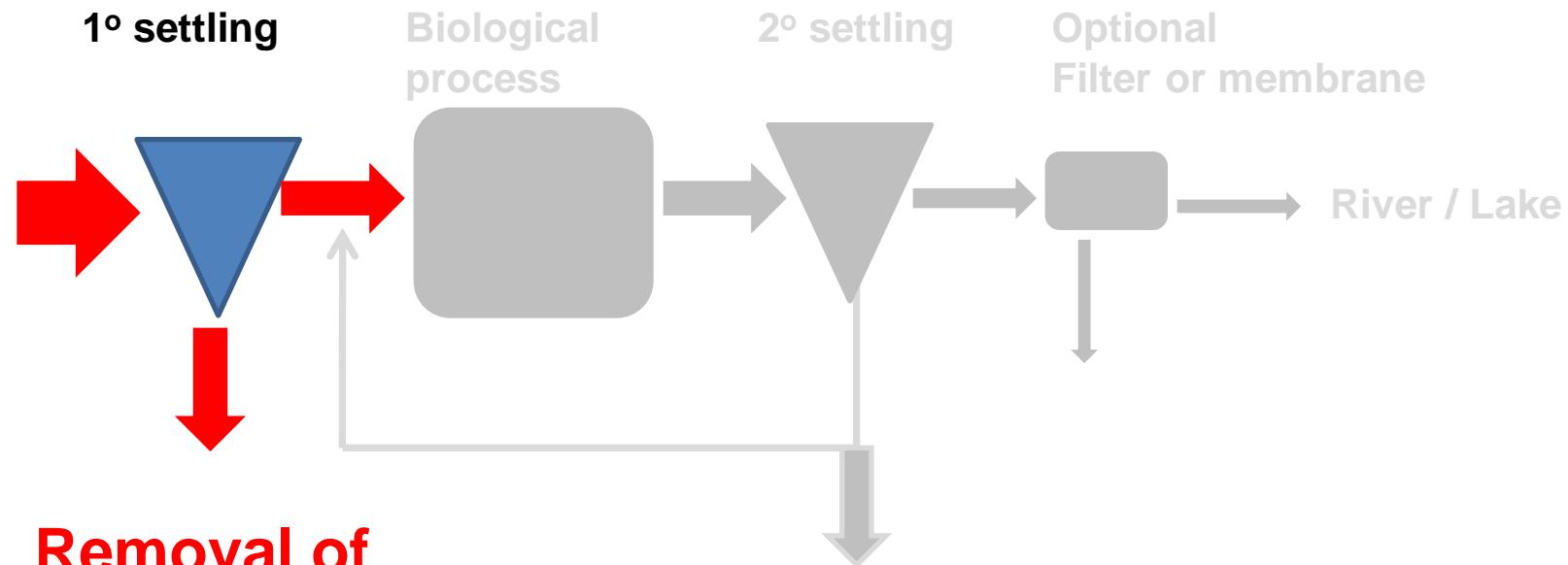


**Red Arrows represent
flux of nanomaterials**





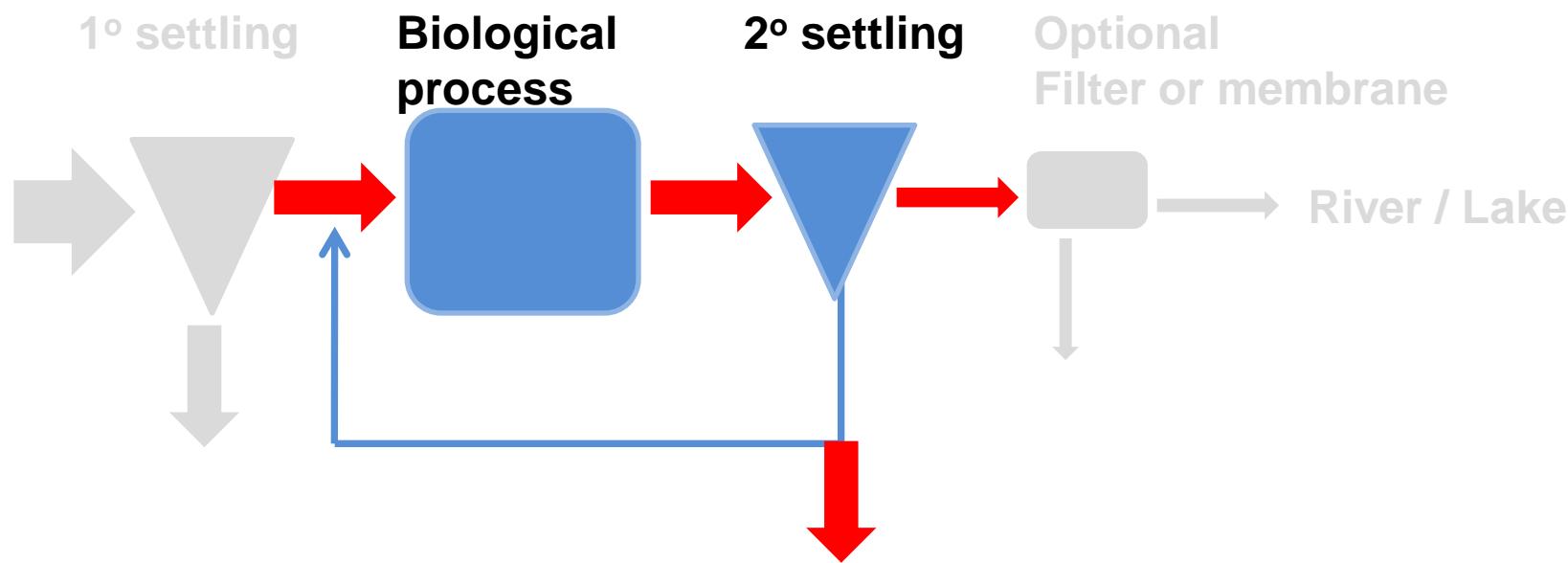
Primary Settling



**Removal of
nanomaterials that are
aggregated to clays,
bacteria or other solids
> 20 um in size**



Biological Treatment



**Removal of
nanomaterials occurs
when they interact with
biofilms or biosolids**



Batch Sorption Experiments

- Fresh wastewater biomass
- Mixing and settling times mimic hydraulic residence times at plant
- Analyze settled supernatant
- Can readily screen many properties
- Quick test
- Standard EPA method exists for organic pollutants using freeze-dried biomass too



Nanoparticle Control
(No Biomass Sorbent)



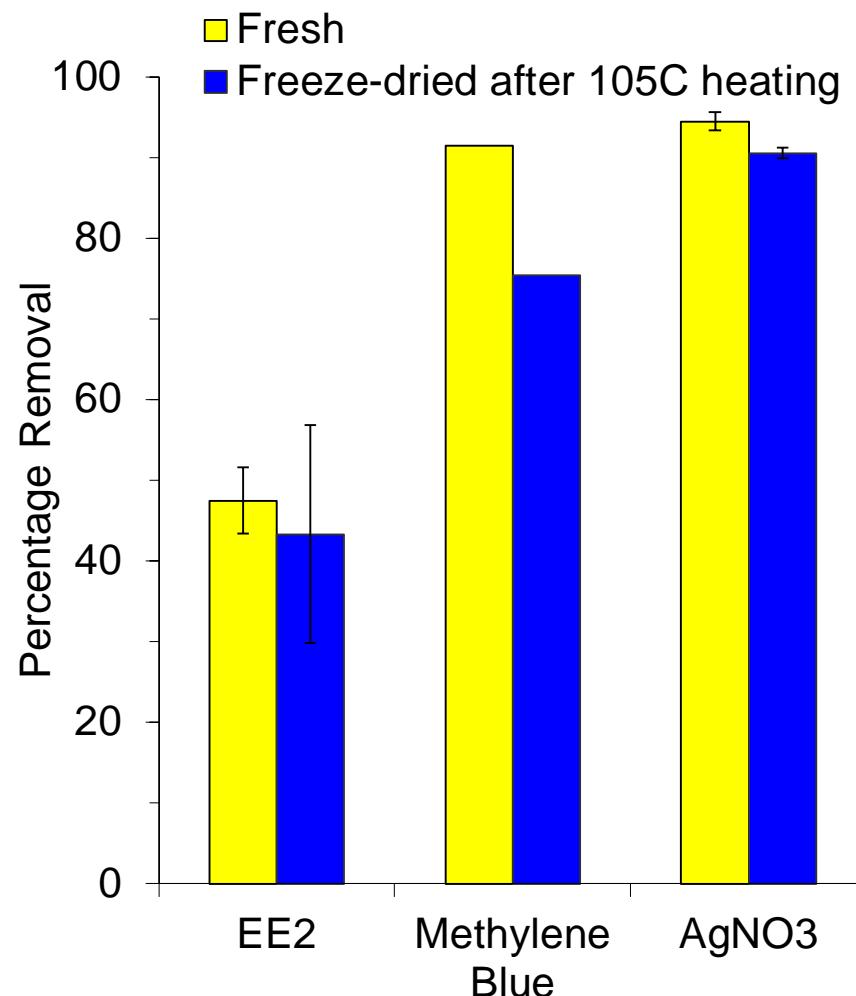
NP + 400 mg TSS/L
Biomass Sorbent



NP + 800 mgTSS/L
Biomass Sorbent

Standard Batch Methods

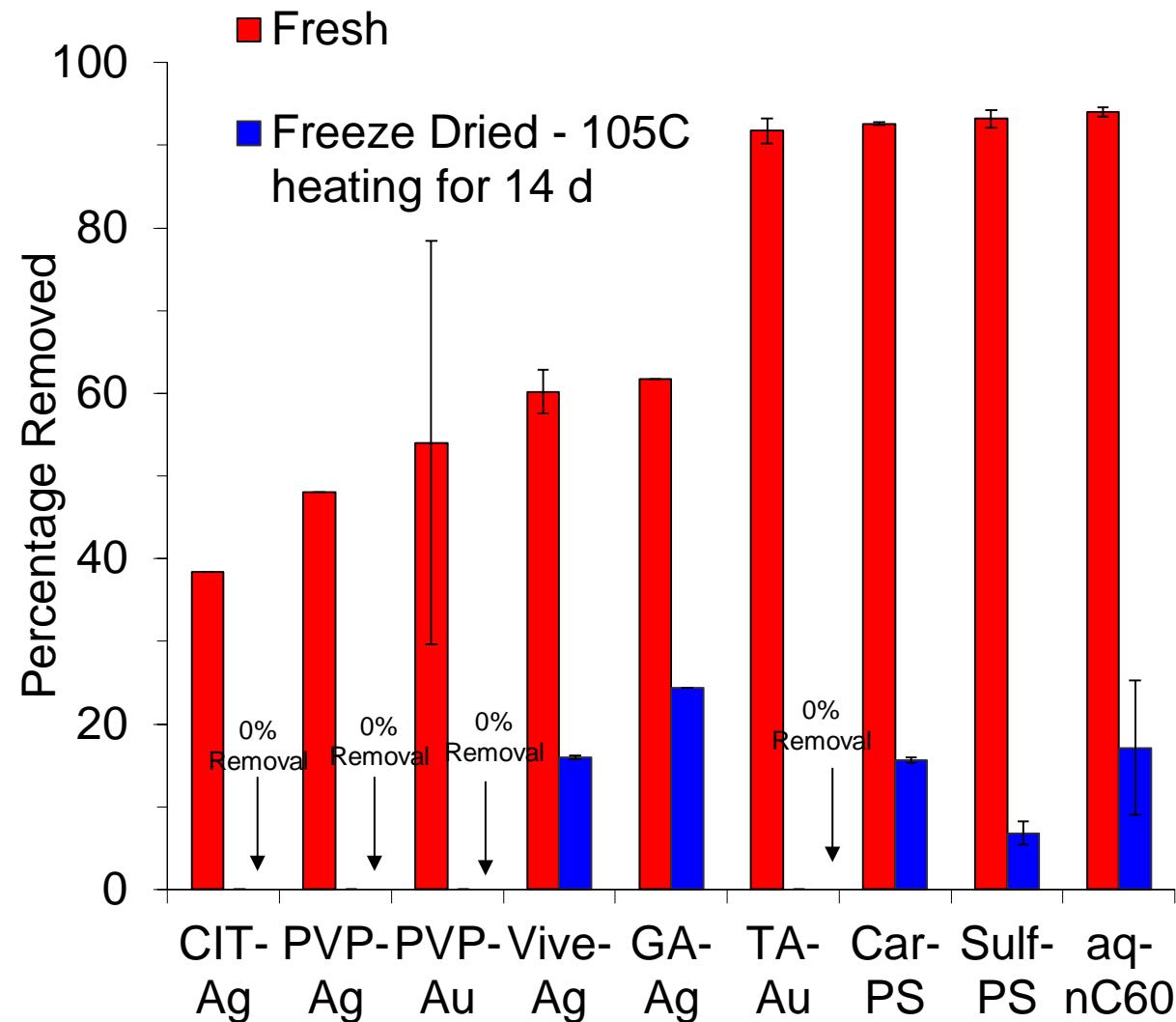
- OPPTS 835.1110
Activated Sludge Sorption Isotherm
- Uses freeze-dried biomass
- Validated for organics, and has been used for metals
- Data here shows fresh and freeze-dried biomass provide comparable removals when applied at similar mgTSS/L biomass





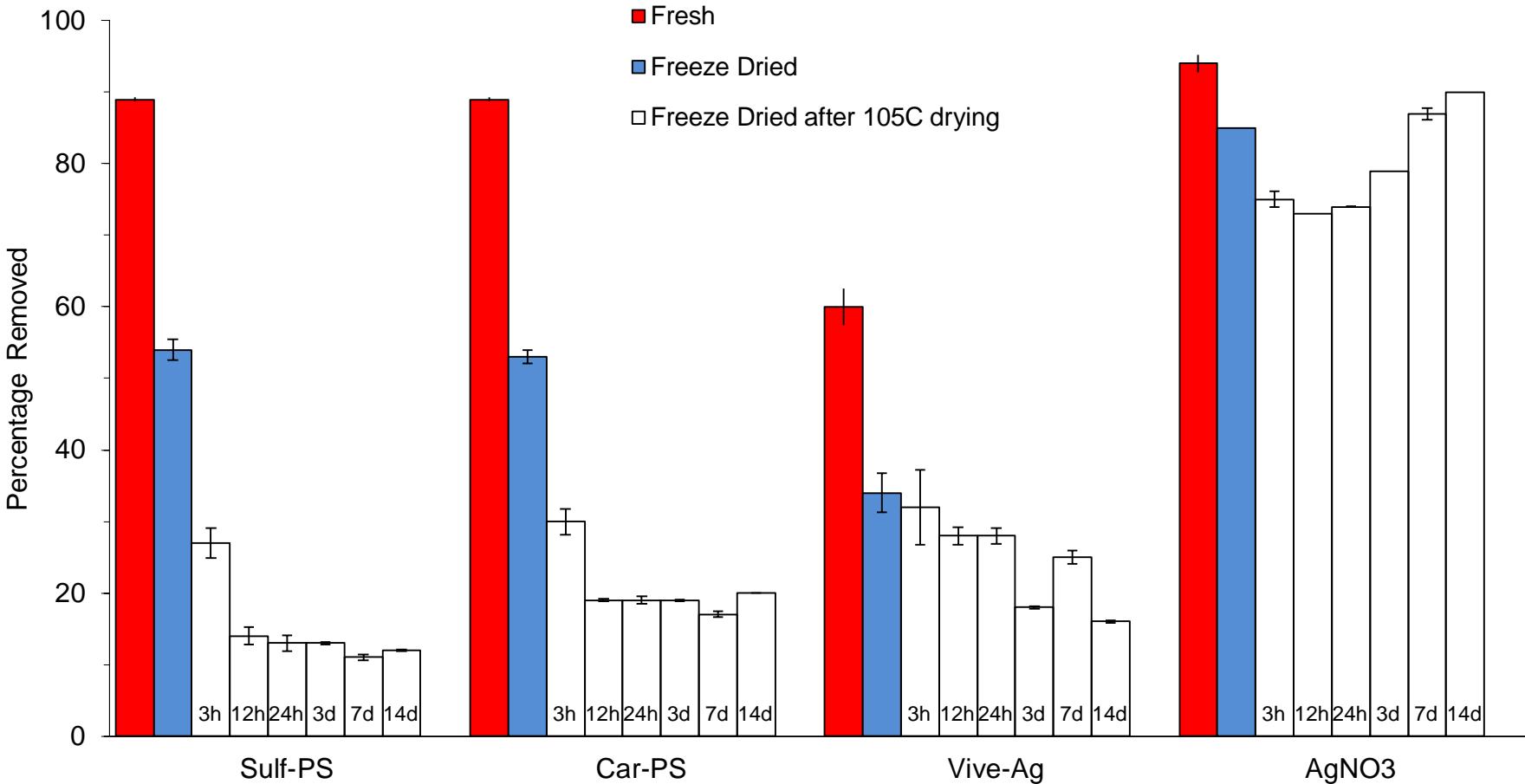
Does OPTT Test work for NPs? NO!

- X-axis is different types of nanoparticles
- Fresh and rinsed biomass shows much more capacity for NMs than freeze-dried biomass



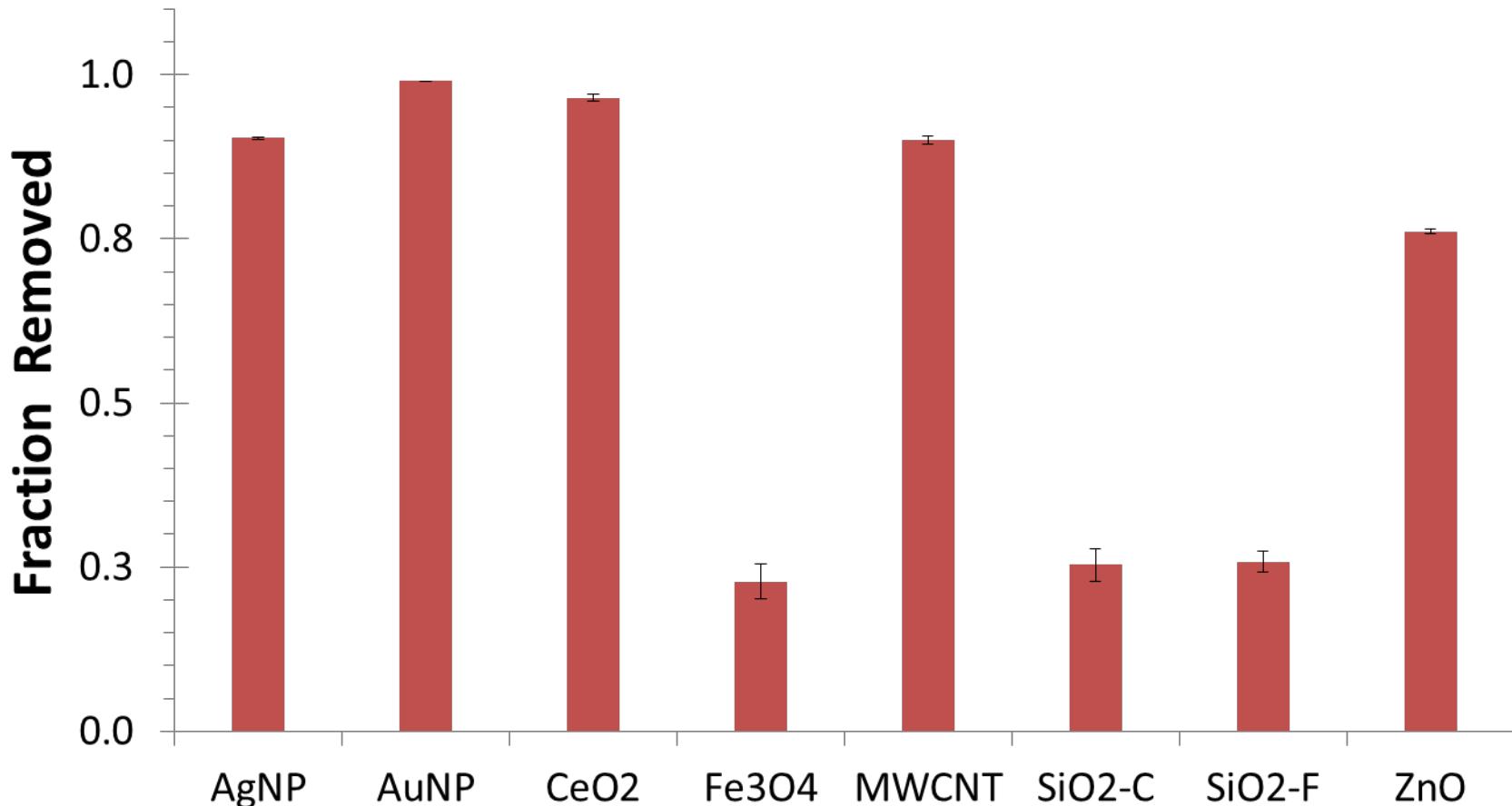


Freeze-drying and heating denatured proteins & released bacterial “surfactants”



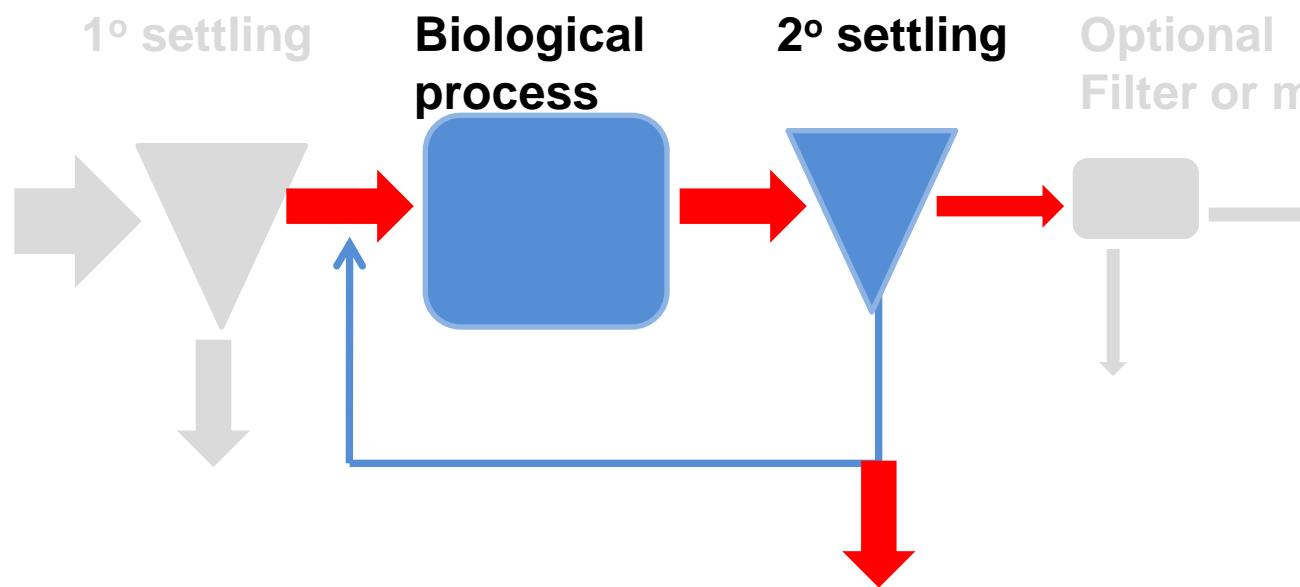


Removal by biomass of different Nanoparticles (1 gTSS/L)

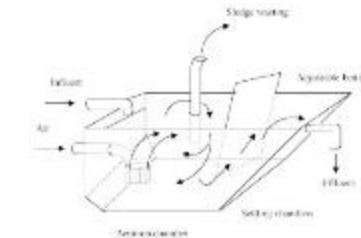




Biological Treatment



Removal of nanomaterials occurs when they interact with biofilms or biosolids



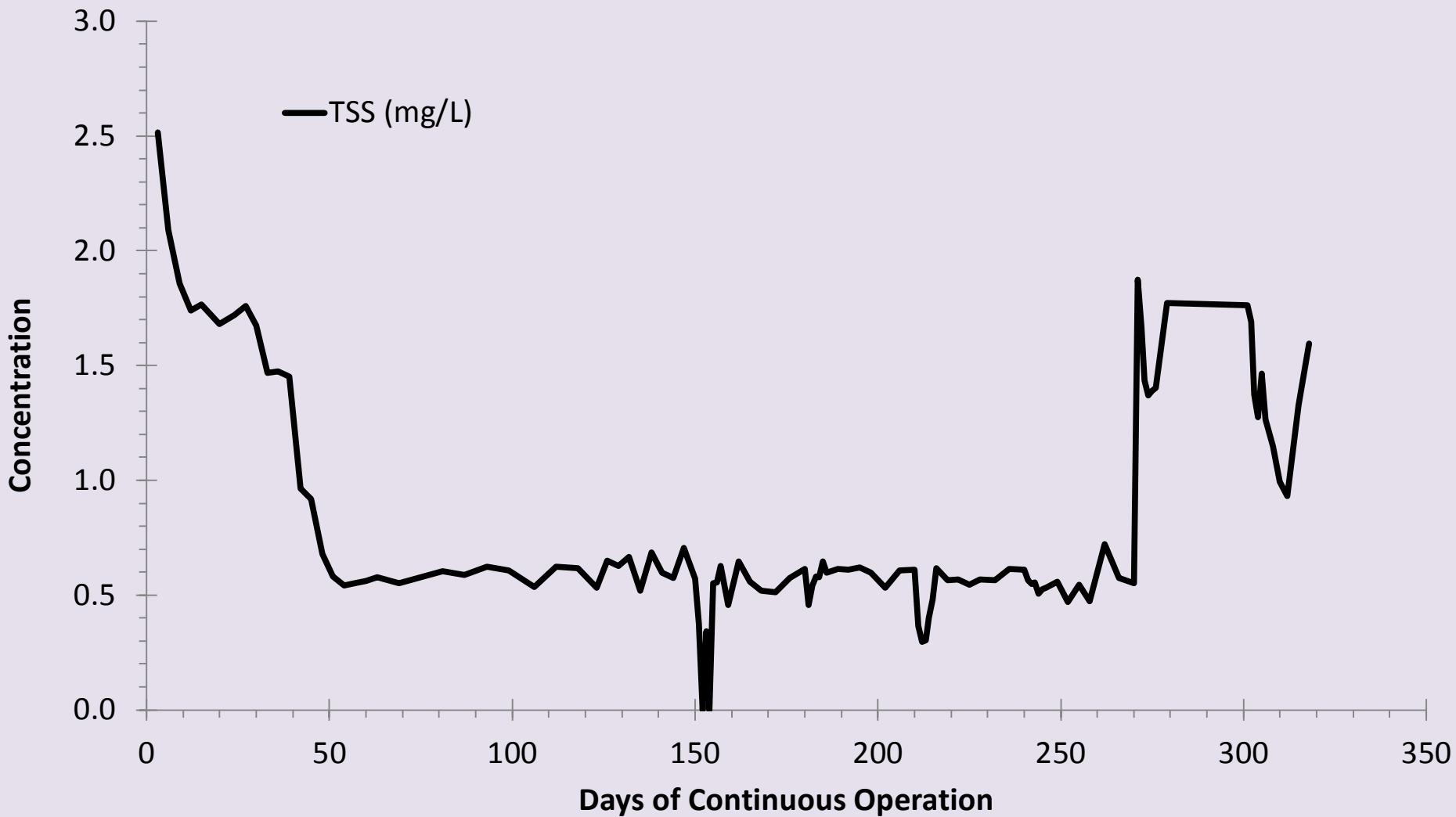
2.5 L
SRT = 6 to 10 days
Influent COD: ~750 mg/L
Influent NP: 0.07 to 2 mg/L



Results of SBR tests with continuous, daily addition of Nanoparticles

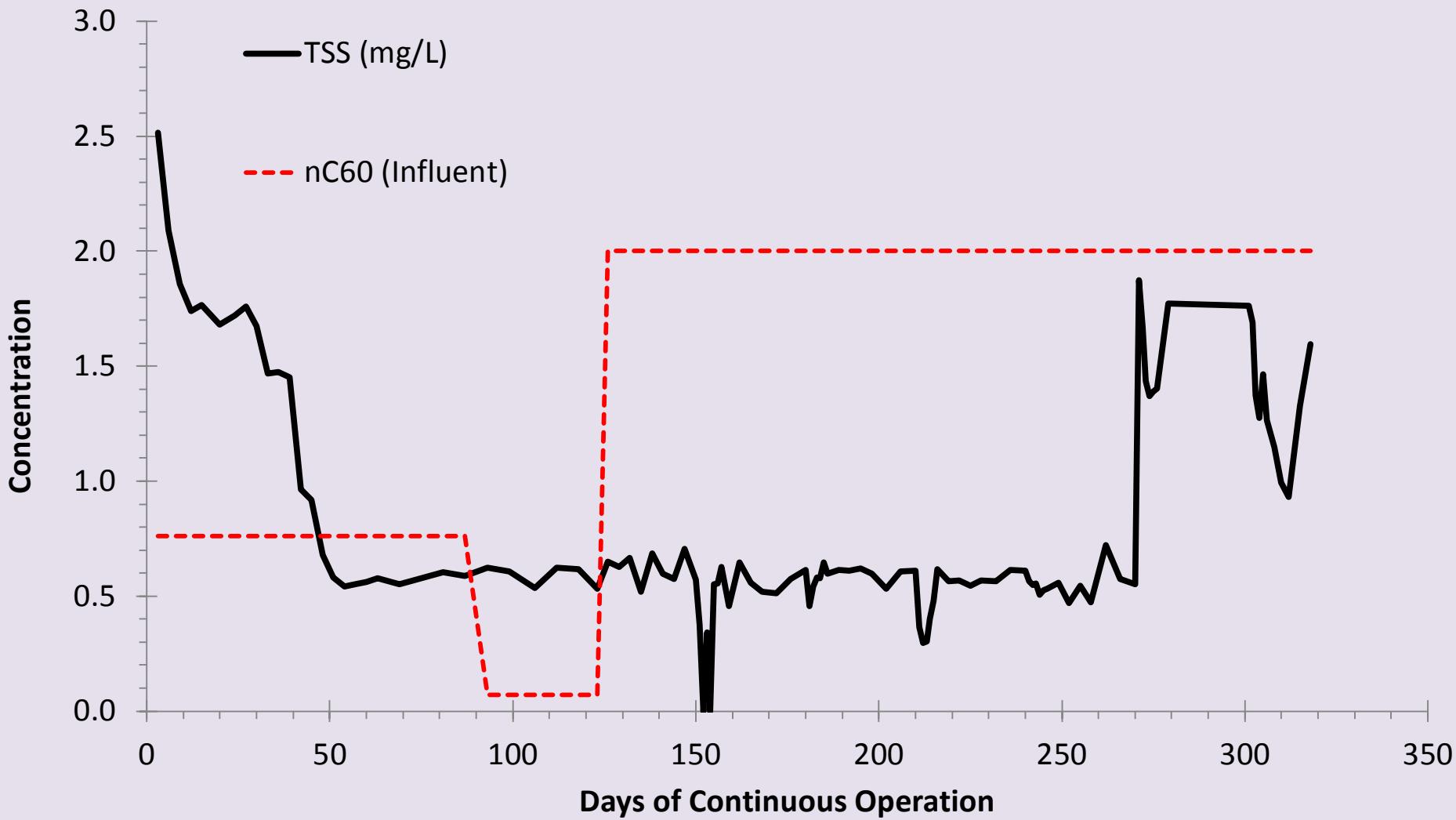


Results of SBR tests with continuous, daily addition of Nanoparticles



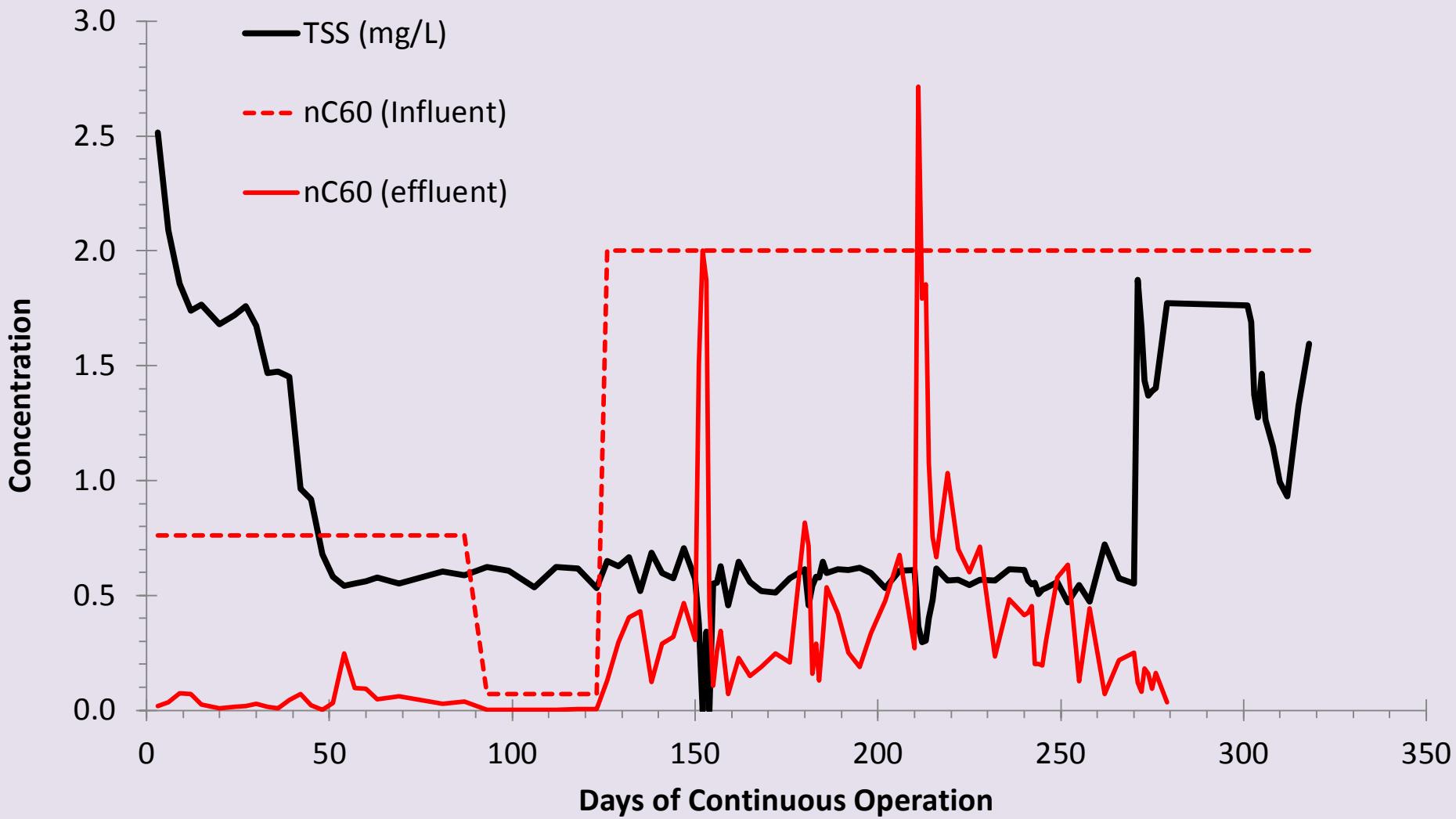


Results of SBR tests with continuous, daily addition of Nanoparticles



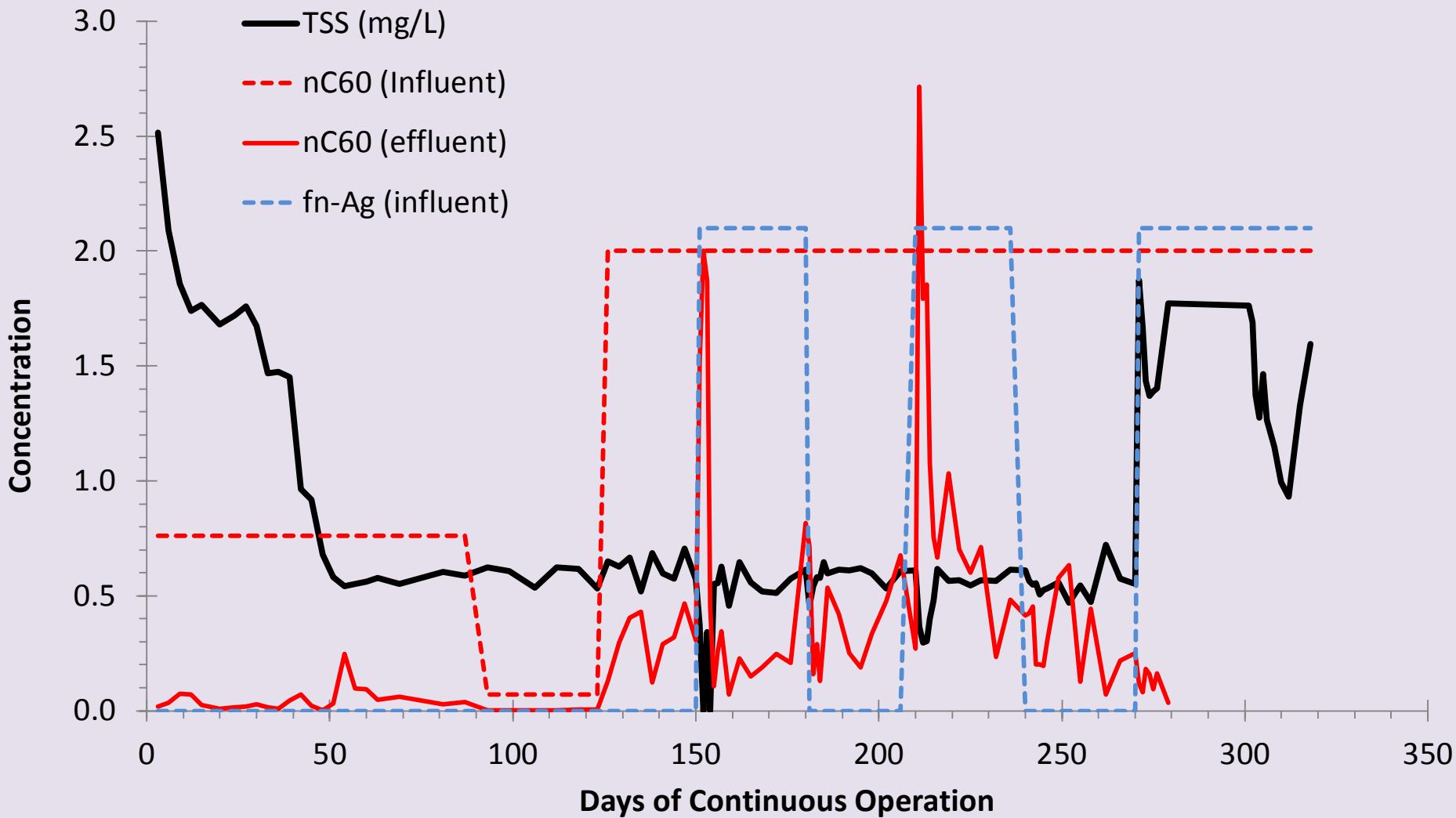


Results of SBR tests with continuous, daily addition of Nanoparticles



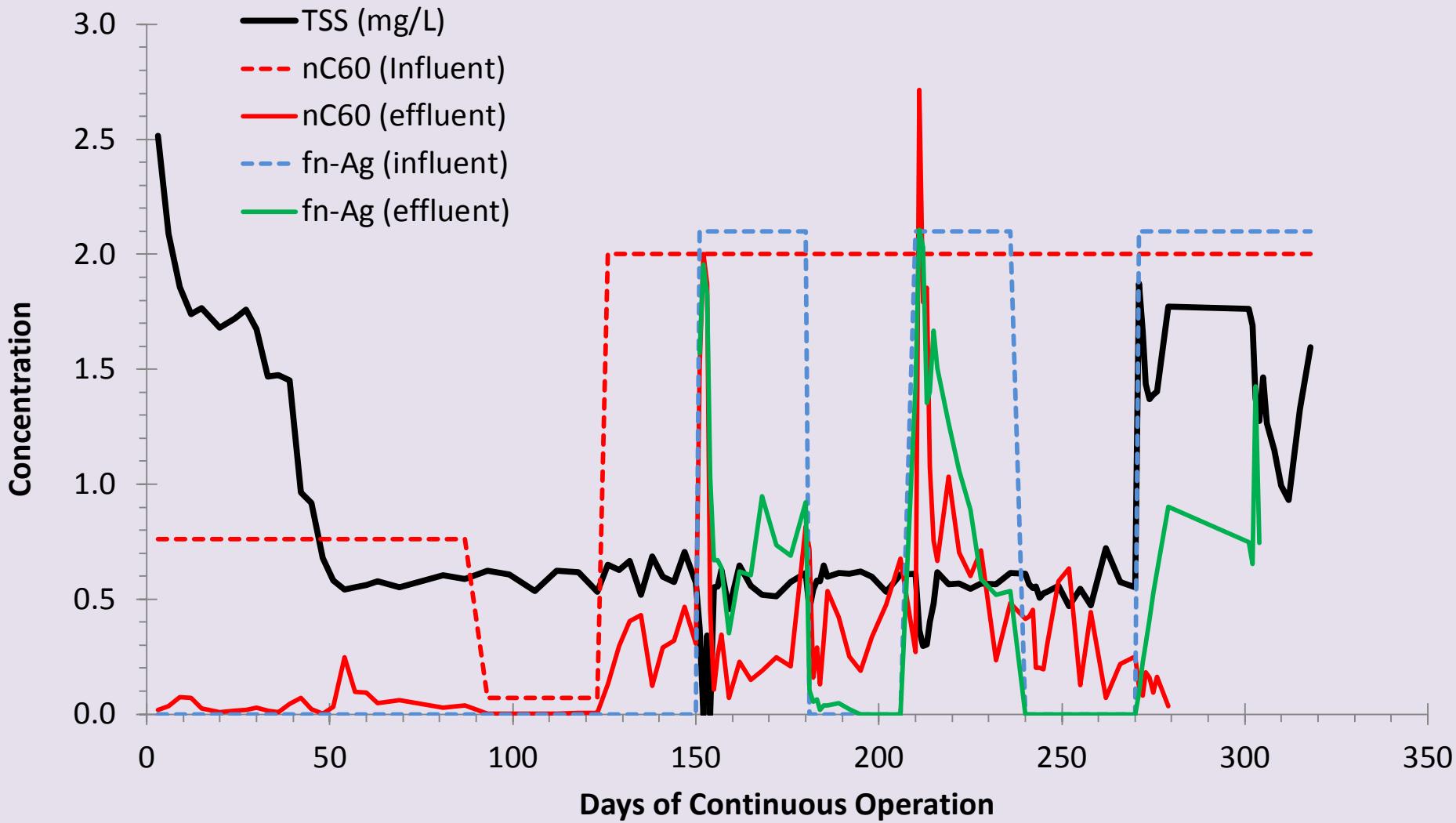


Results of SBR tests with continuous, daily addition of Nanoparticles





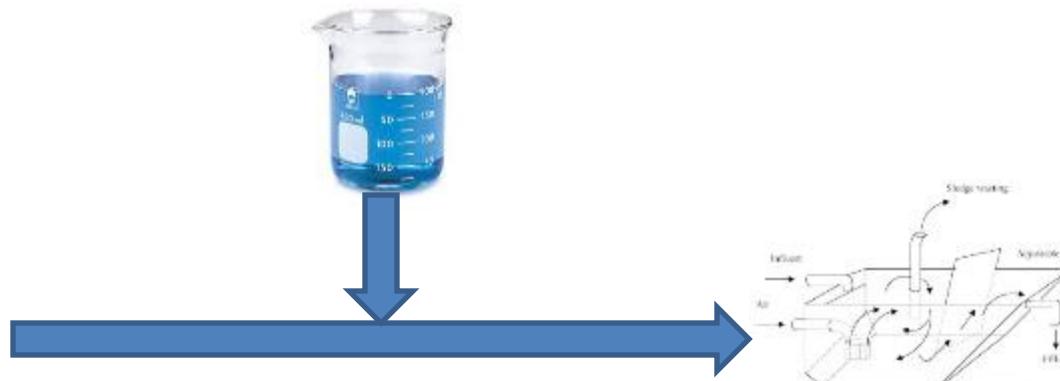
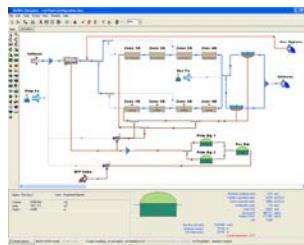
Results of SBR tests with continuous, daily addition of Nanoparticles





Modeling NP Removal

Kiser et al.,
Sep. Sci Tech. 2010



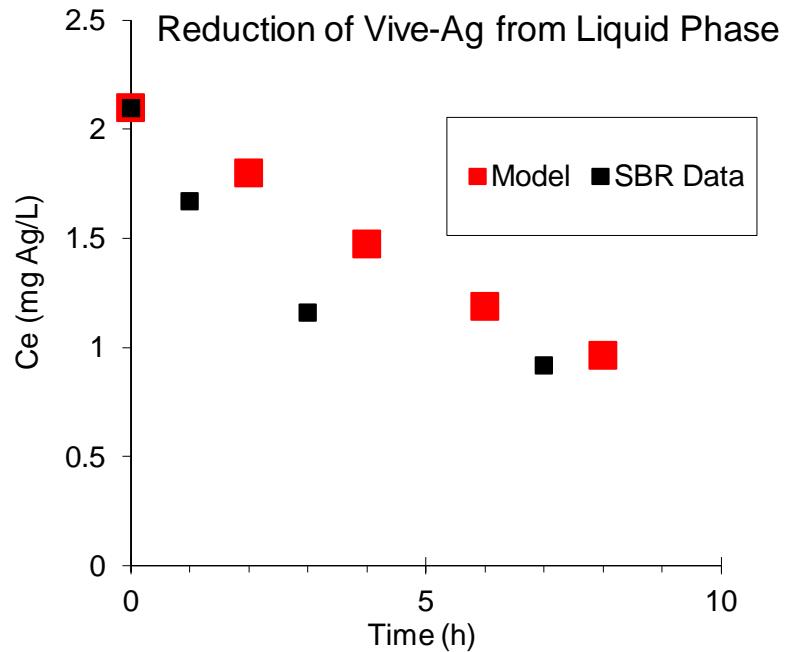
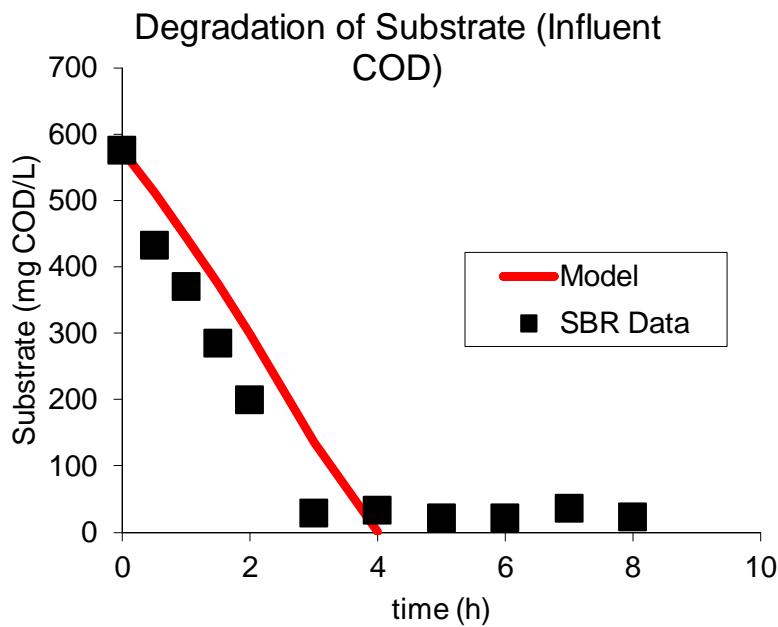
Hypothesis: Batch NP sorption experiments linked with dynamic bacterial growth models and reactor models can predict nanomaterial removal

Example:
Isotherm in batch reactor
Fresh biomass
10 nm diameter citrate functionalized nano Silver
Linear Partition Coefficient for : $K = 0.0144 \text{ L/g}$



Model Predictions

“Isotherms” with NPs predict removal in Sequencing Batch Reactors

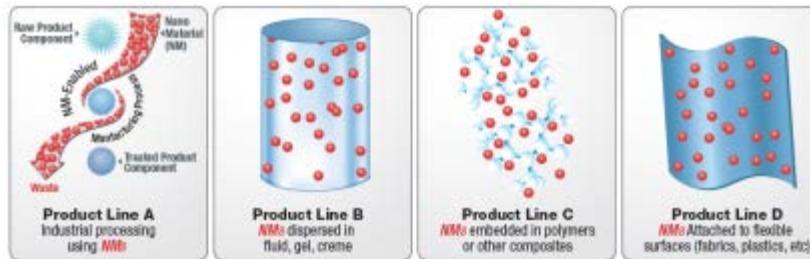


Conclusion: Preliminary confirmation exists that we can go from batch experiments to simulations of continuous flow performance. Difference in NP effluent concentration probably related to NP association with non-settling colloids



Part I - Summary

- Products potential to release NMs into the environment can be grouped into 4 product lines
- NMs are released from commercial products into sewage wastewater



NM Sources & Uses in Society

Wastewater Treatment Plants

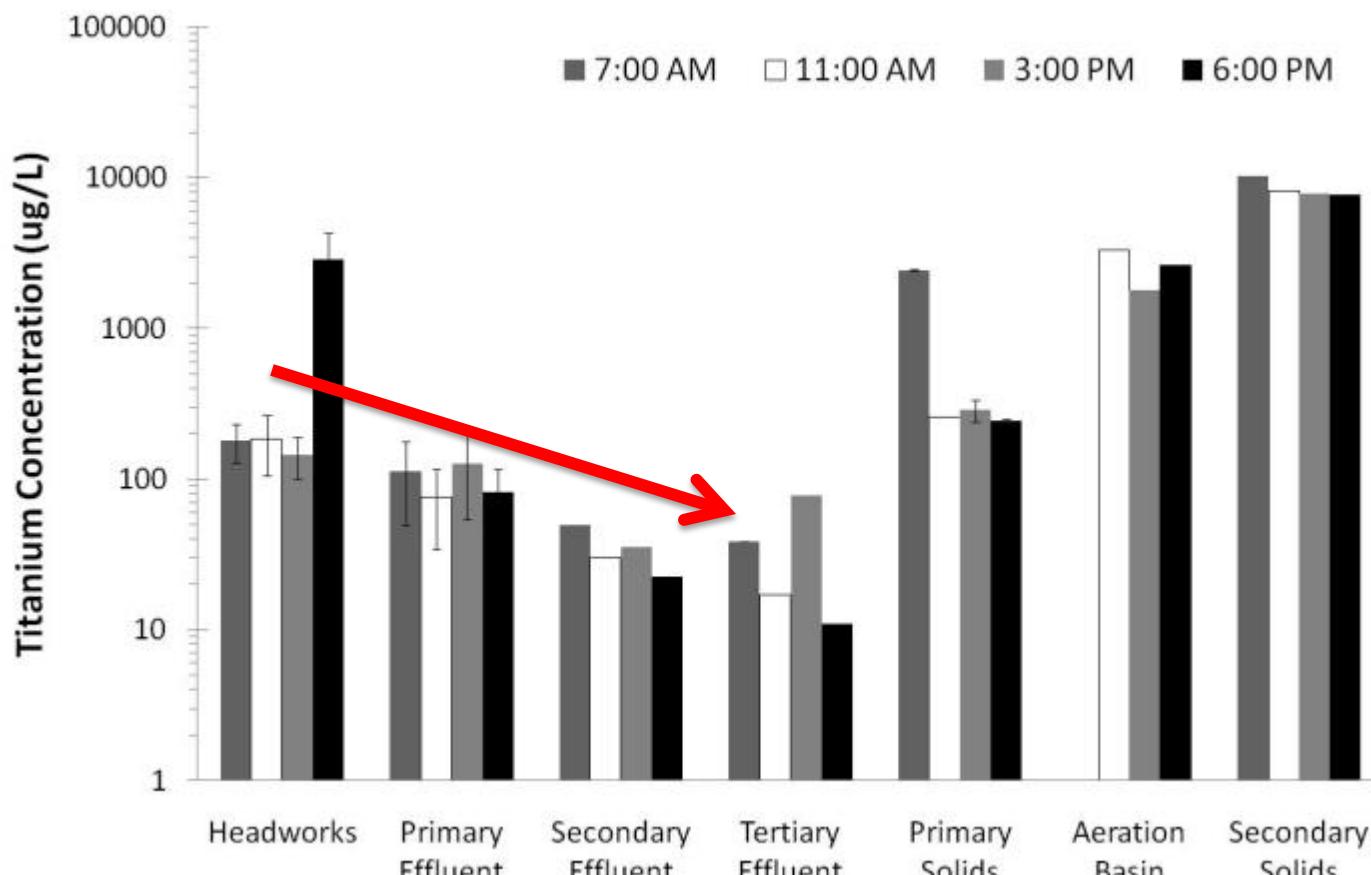
Liquid Effluent to Surface Waters

Biosolids to Land Application, etc



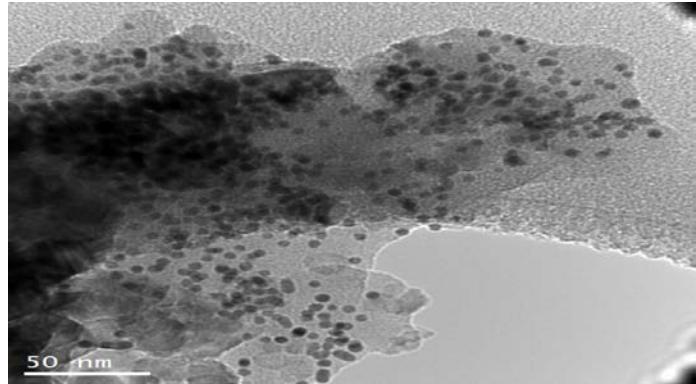
Titanium at Full Scale WWTPs

Titanium is removed from liquid & accumulates in biomass

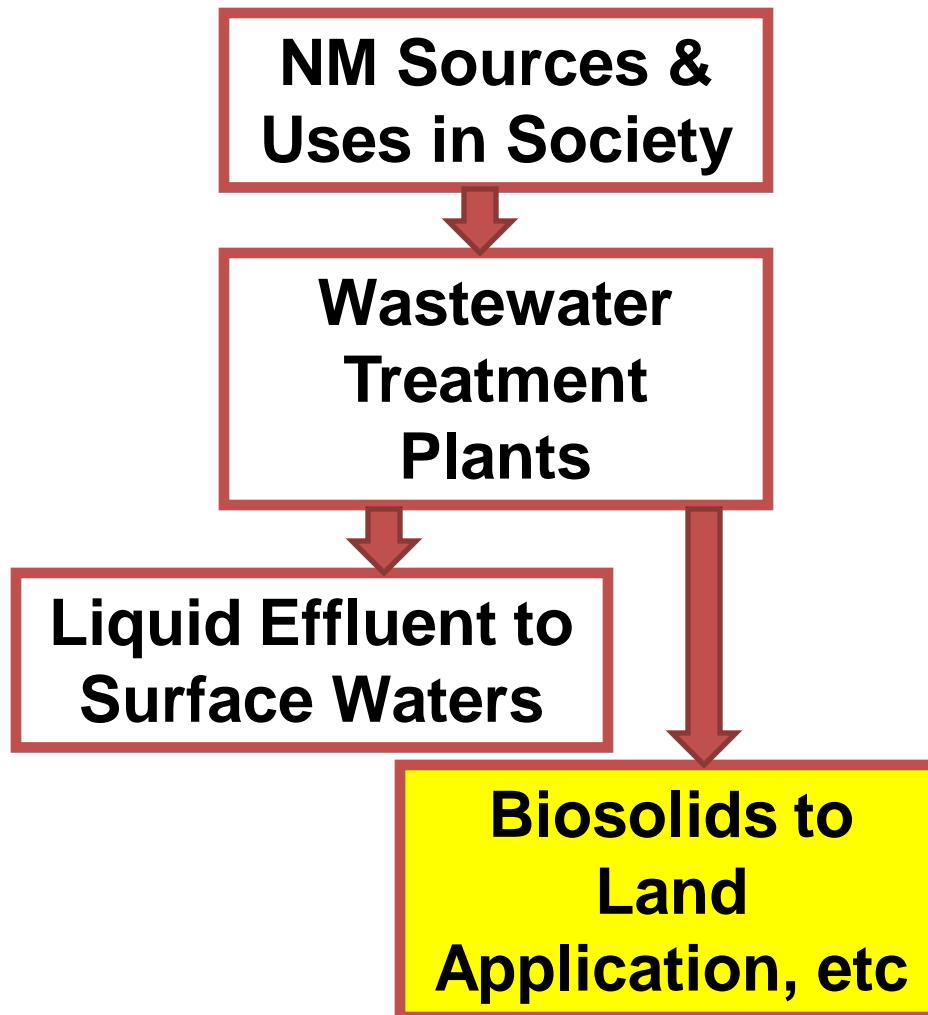




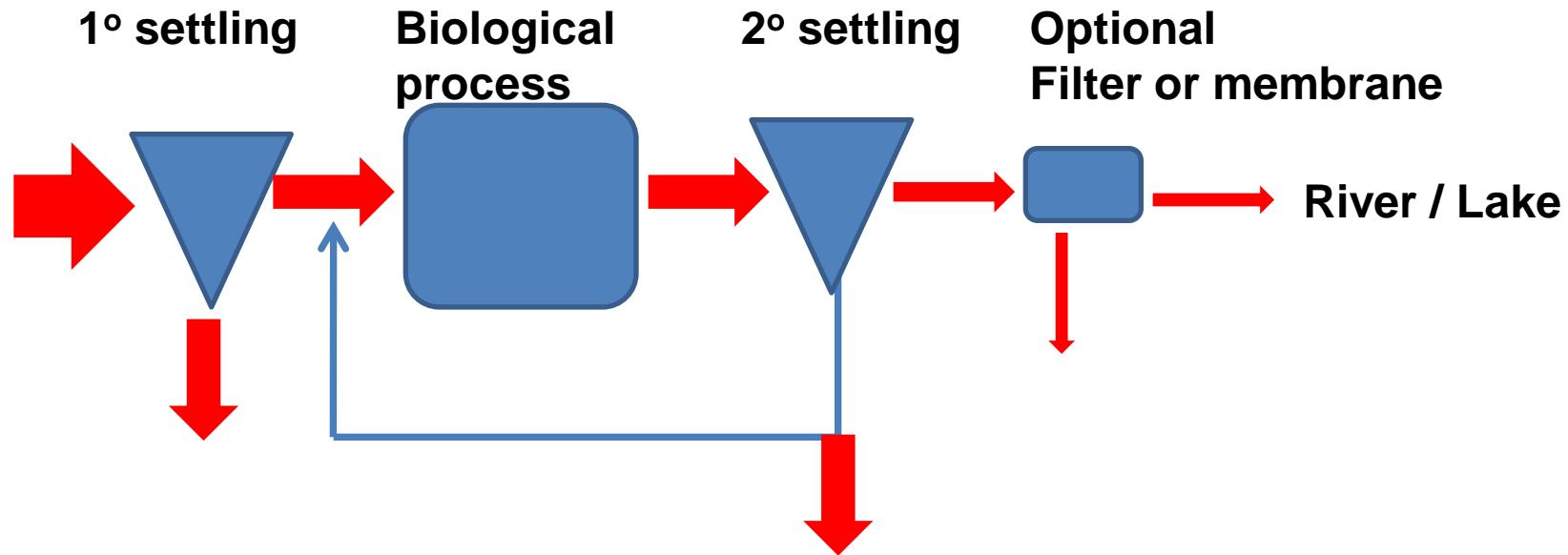
Titanium at Full Scale WWTPs



Different Facilities	Titanium Content of water (ugTi/L)	
	Headworks	Effluent
Activated sludge	615	5
Act. Sludge + filter	180	7
Activated sludge	363	3
Activated sludge	141	2
Activated sludge	581	18
Activated sludge		8
Activated sludge	233	2
Trickling filter	549	13
Membrane bioreactor (MBR)	310	1
MBR	422	4
Average	377	6



Under typical TSS levels (1500-3000 mg/L) greater than 90% of even highly negatively charged NMs will *distribute* into biomass



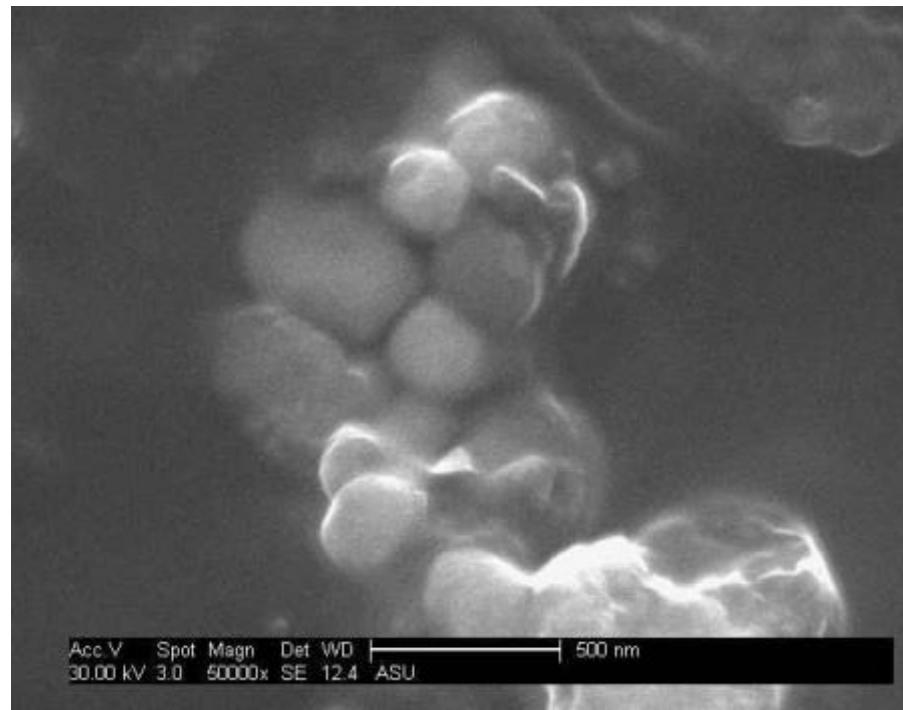
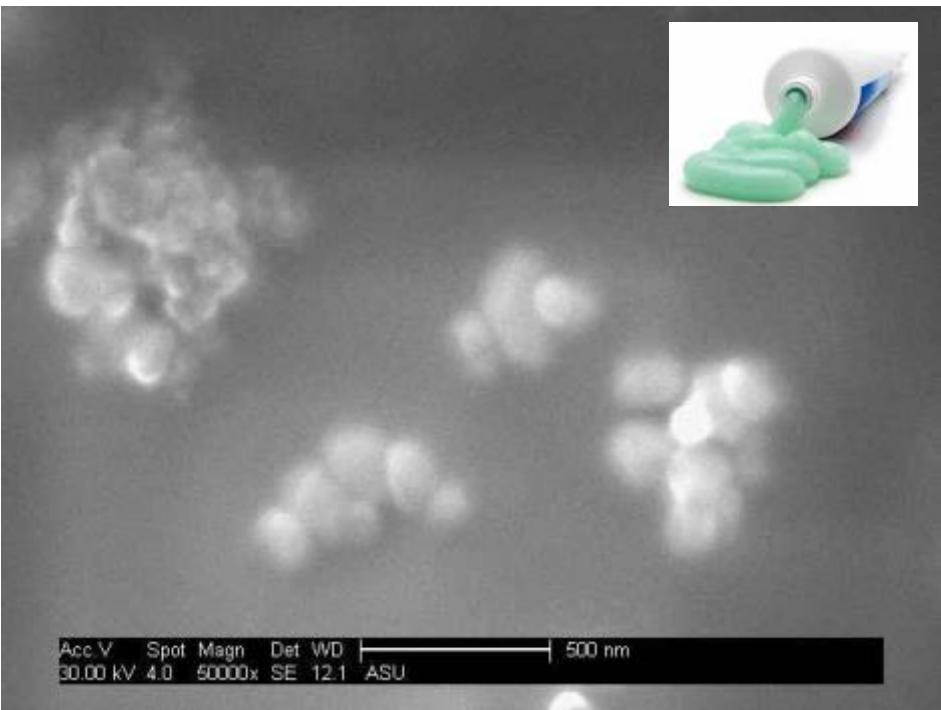
*Let's go
Nanoprospecting*



What type of NMs do we find in real biosolids?



TiO₂ in commercial products are similar to TiO₂ extracted from biosolids



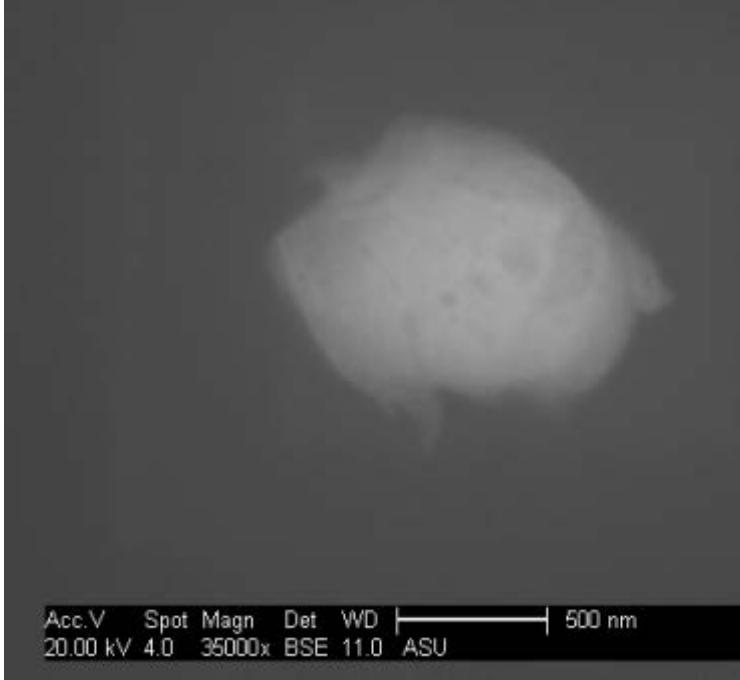
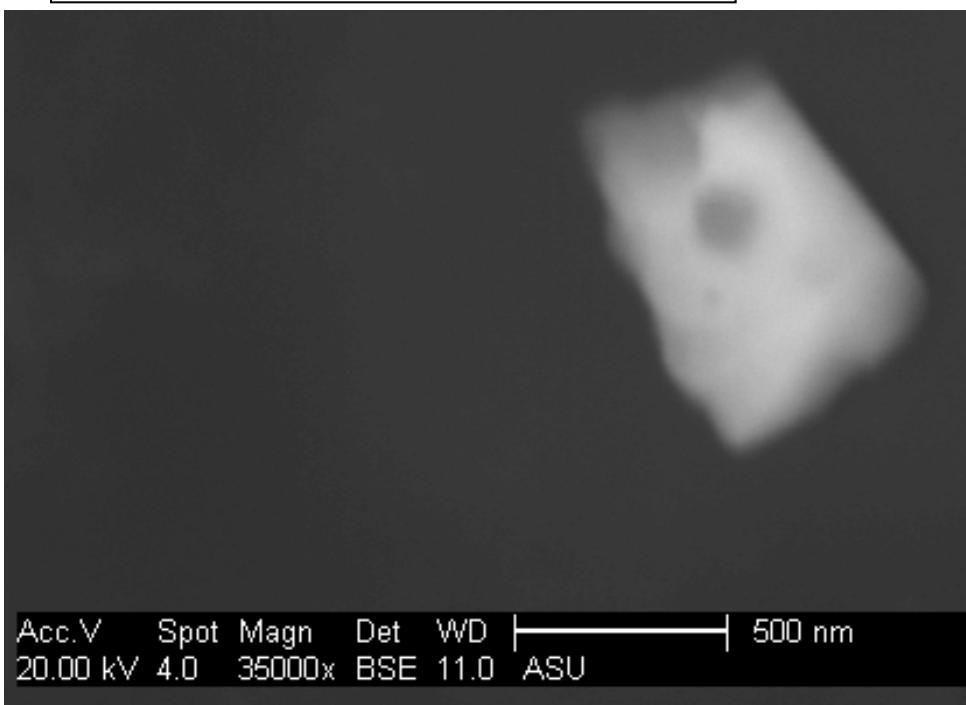
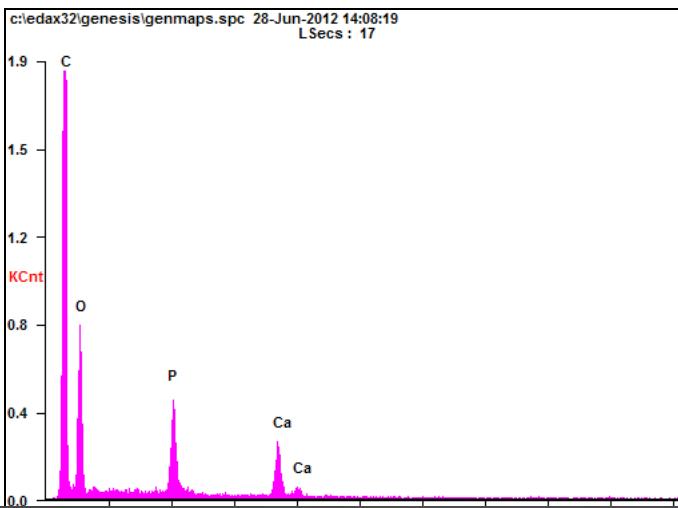
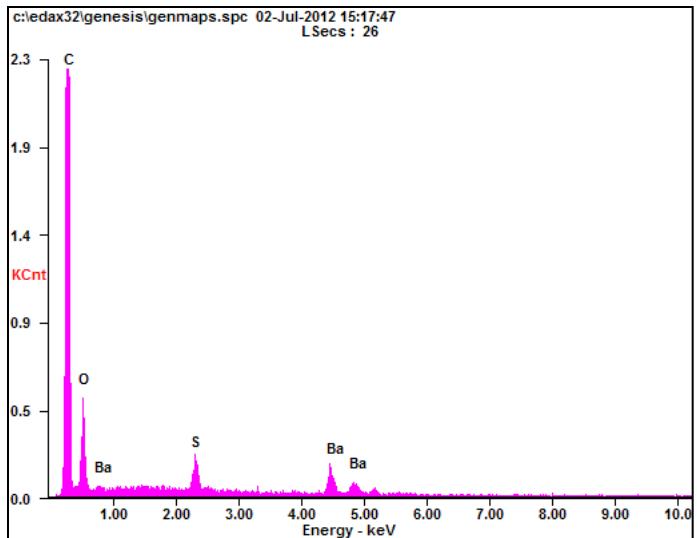
TiO₂ in Toothpaste

TiO₂ in Biosolids



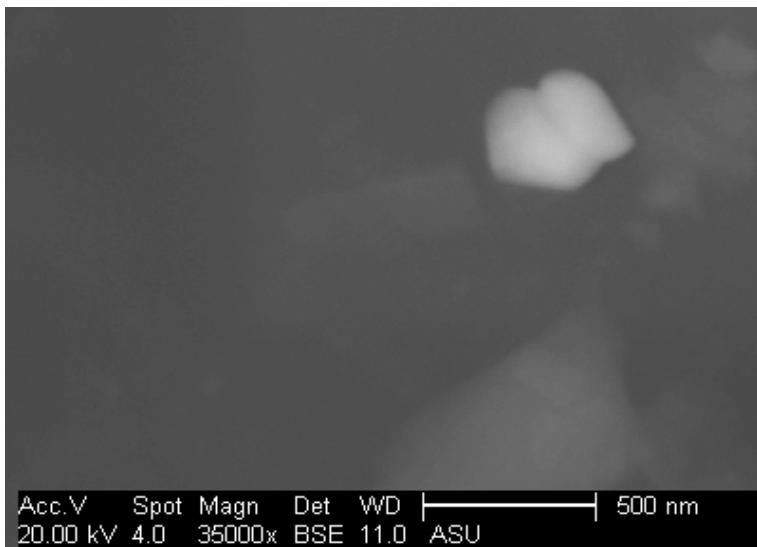
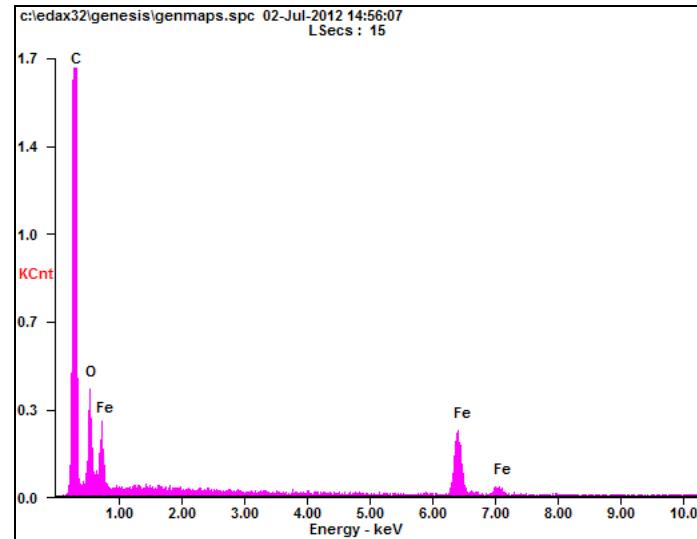
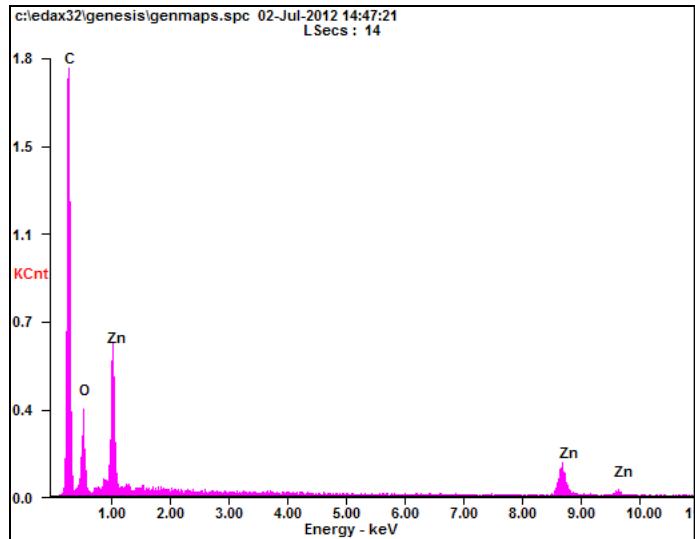
Barium Sulfate & Calcium Phosphates

ASU



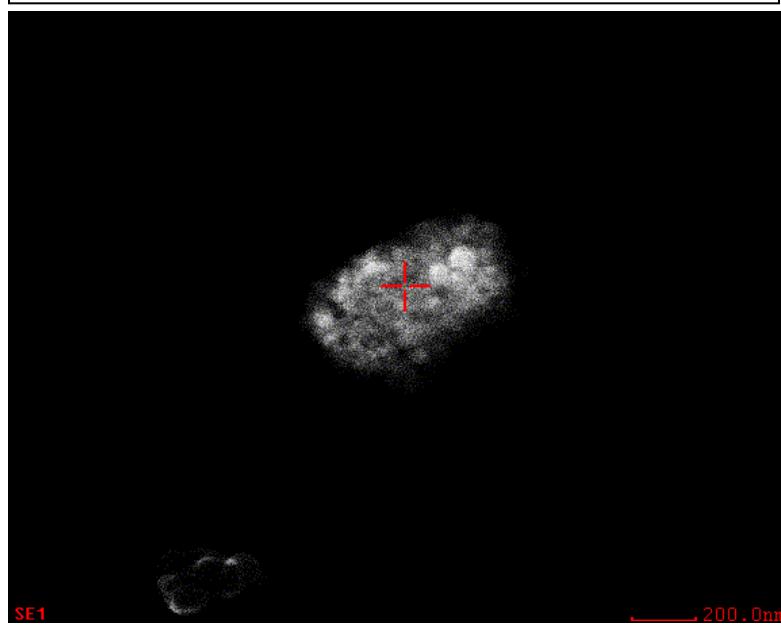
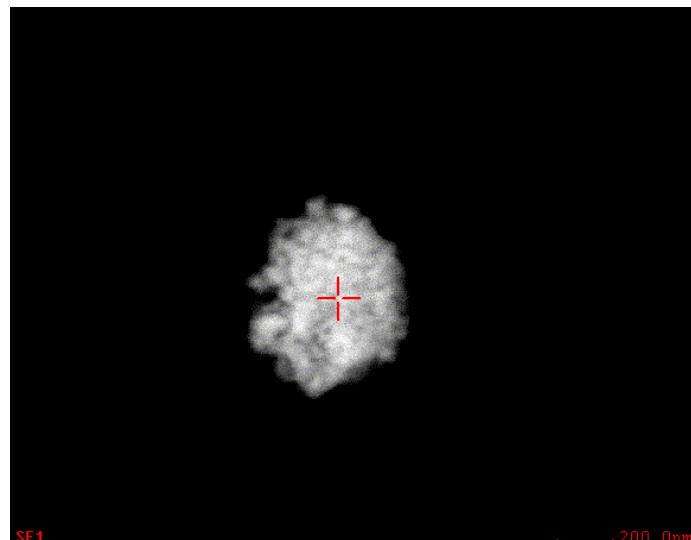
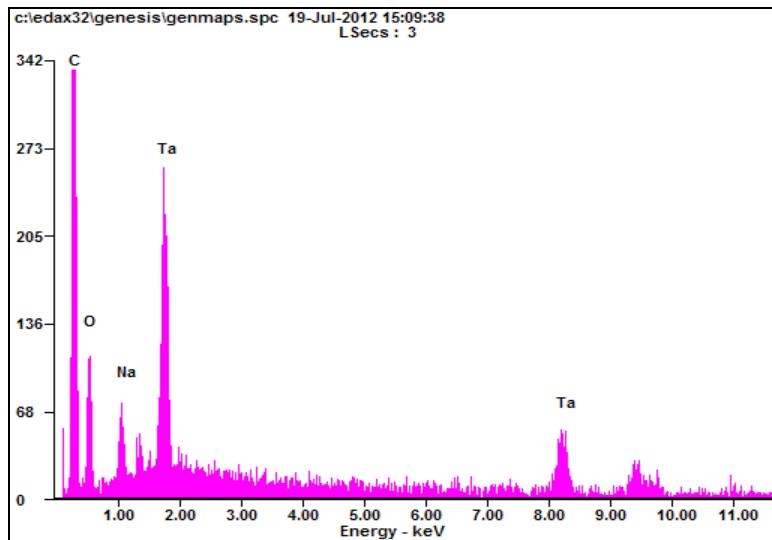


Zinc & Iron Oxides



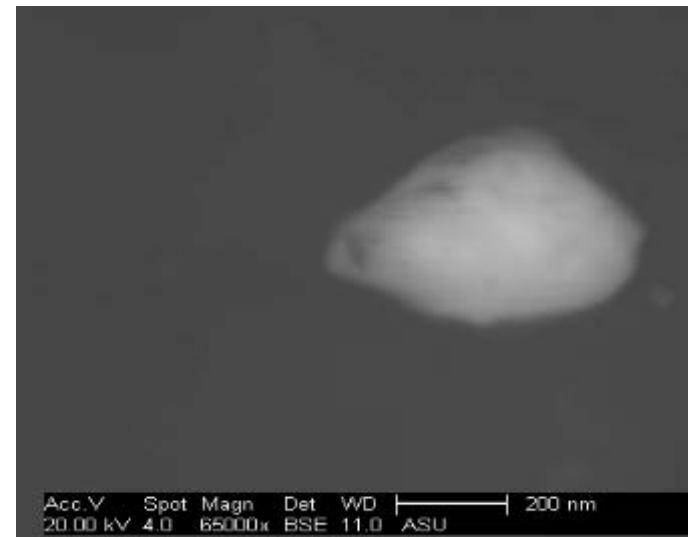
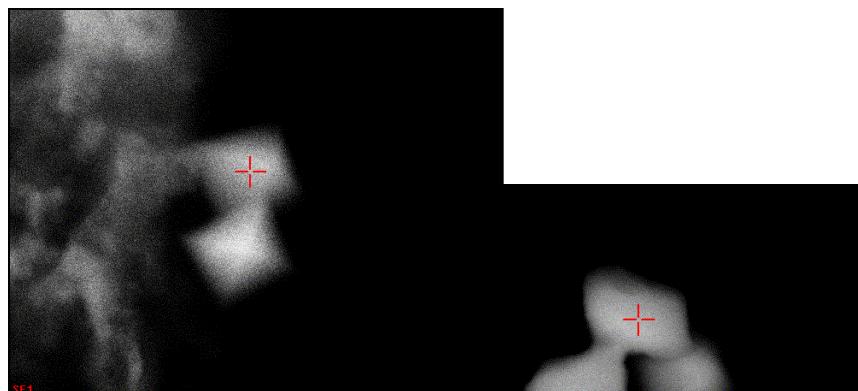
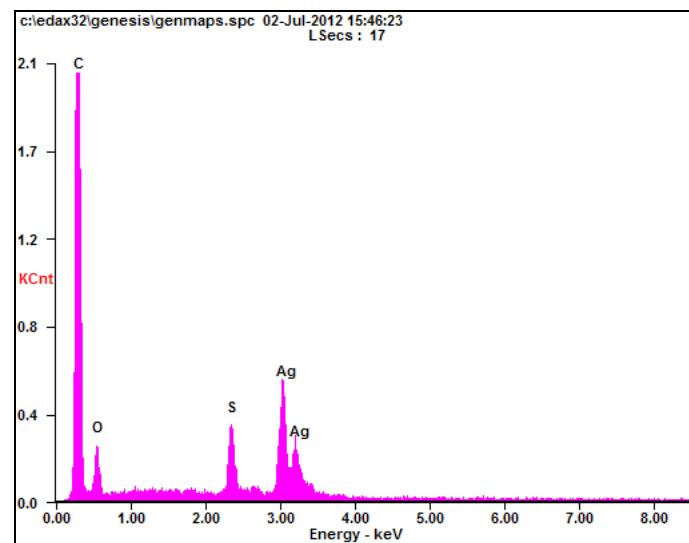
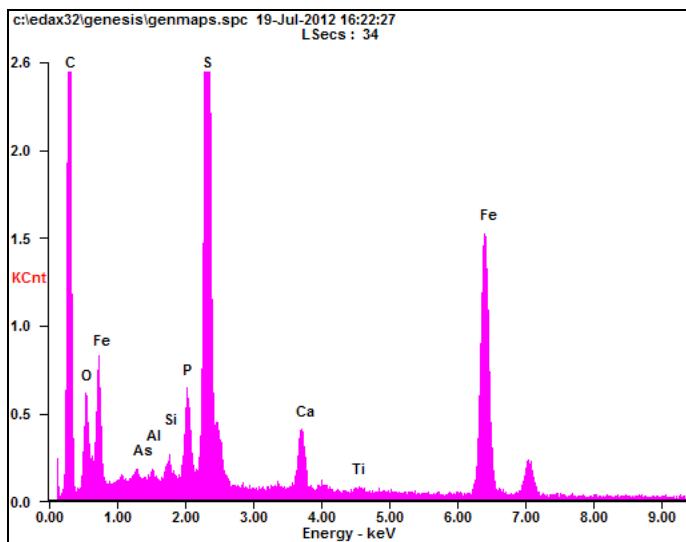


Sodium Tantalate Catalysts





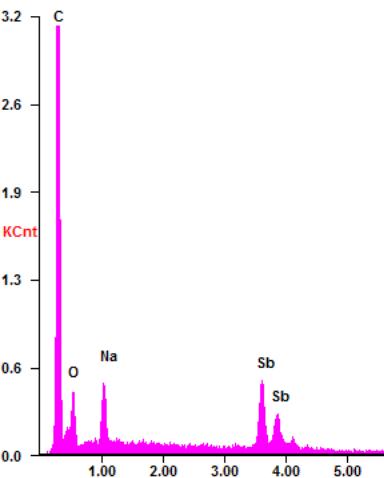
Iron & Silver Sulfides



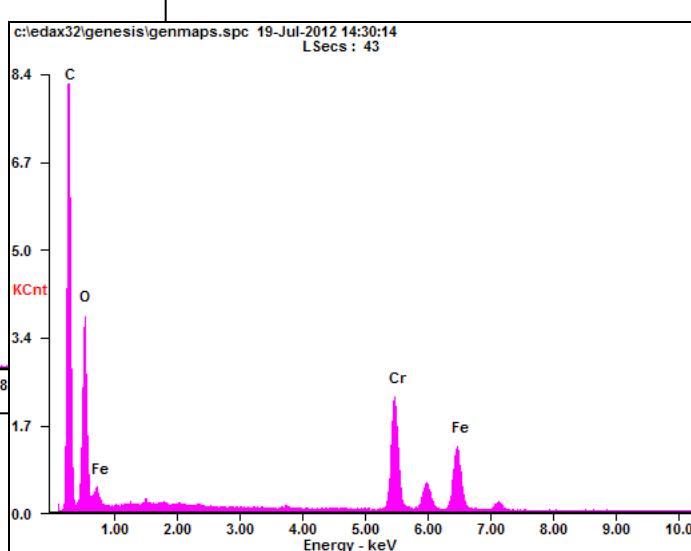


Solder & Facet “released” material

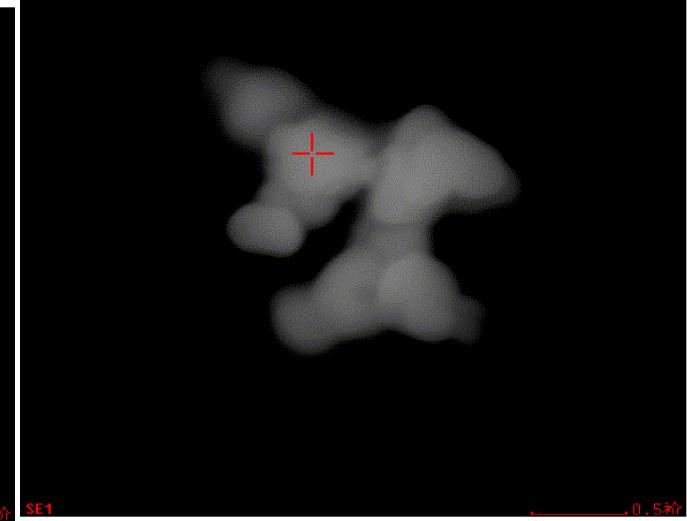
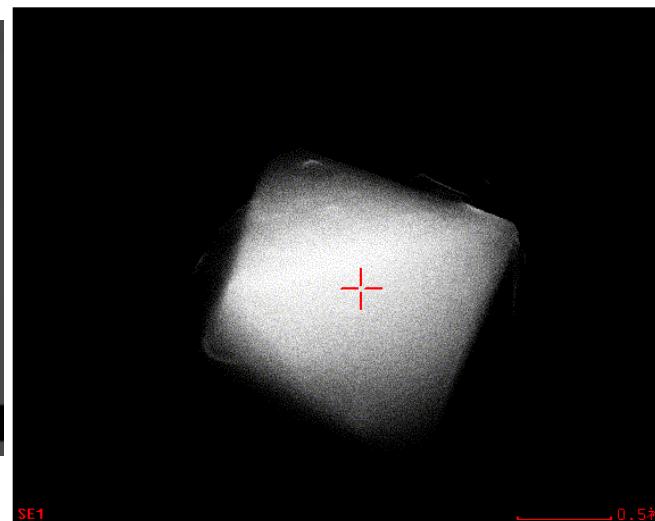
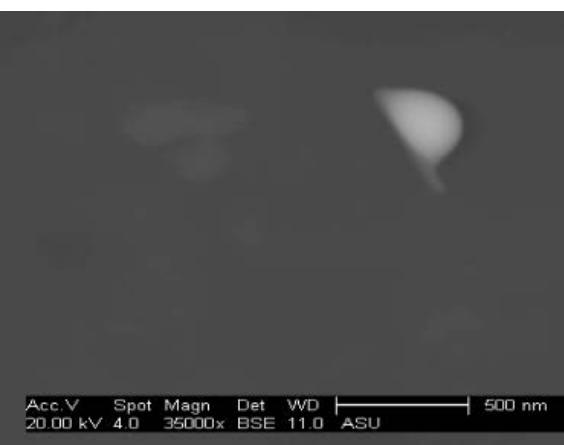
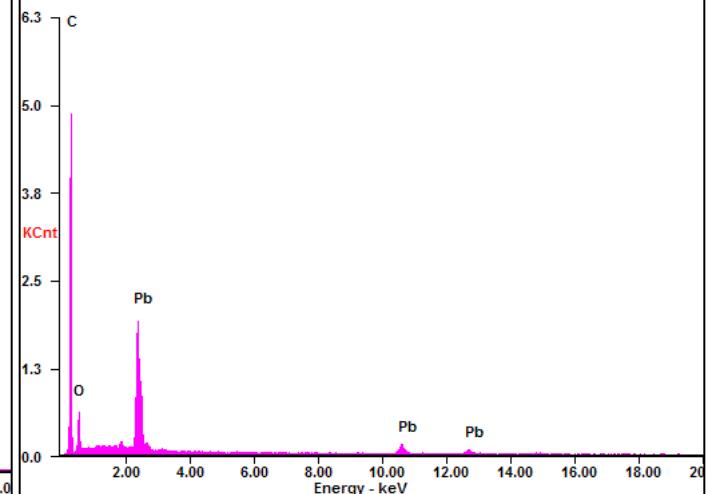
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LSecs : 40



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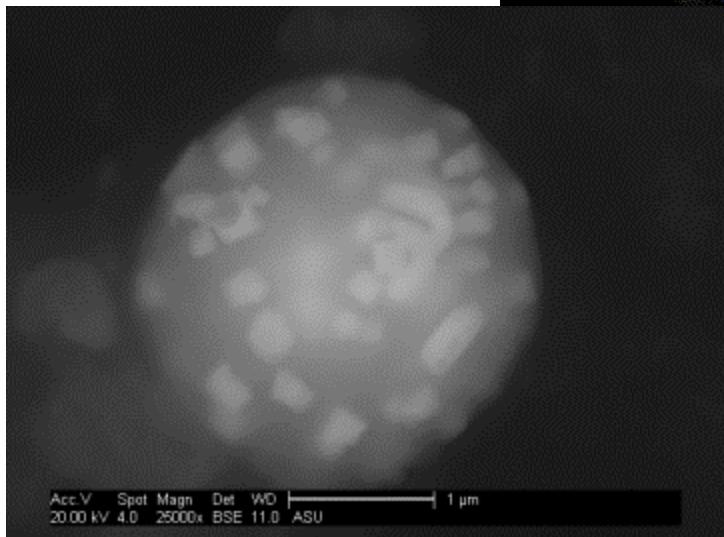
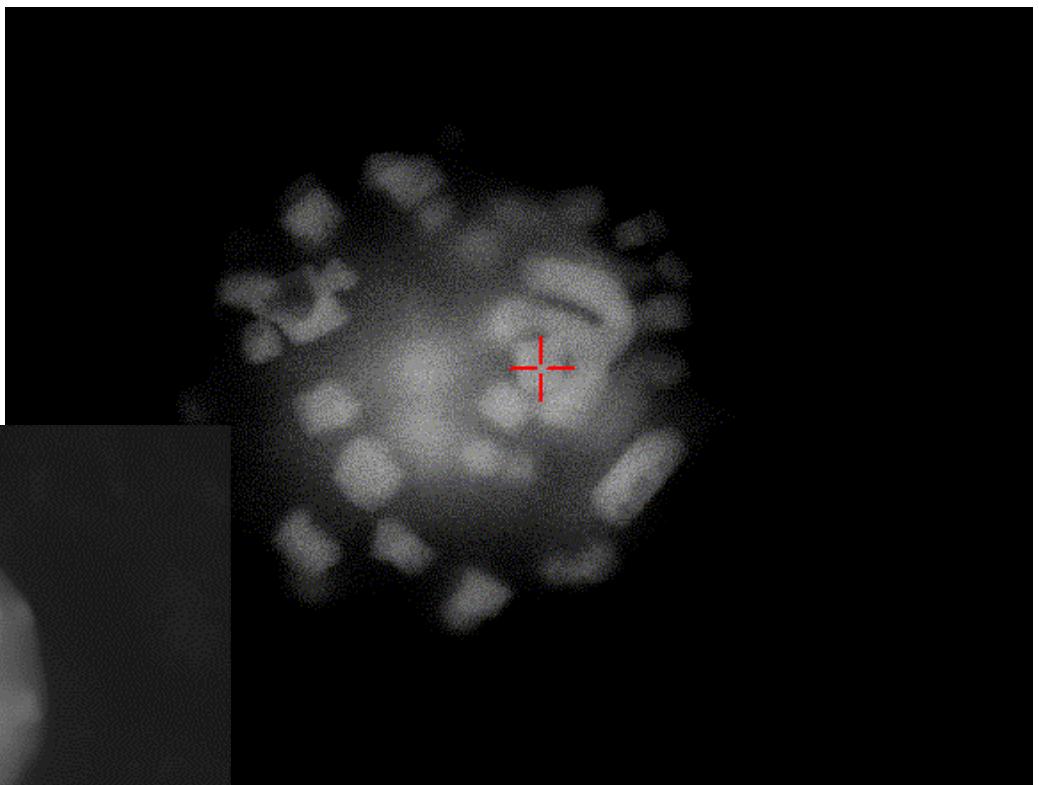
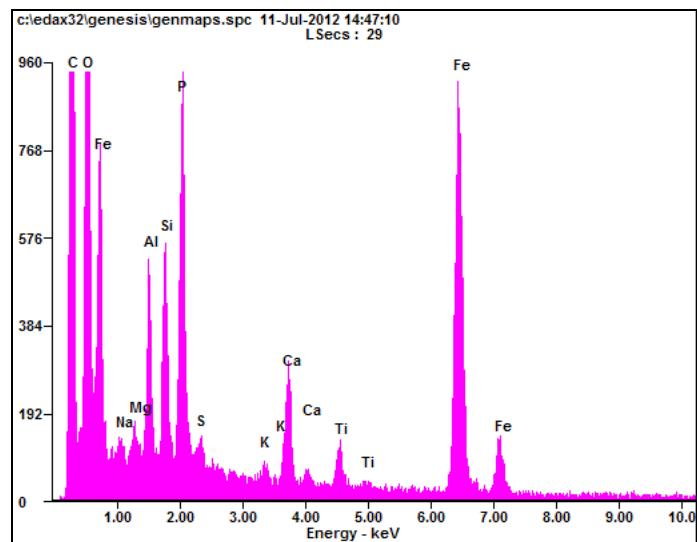


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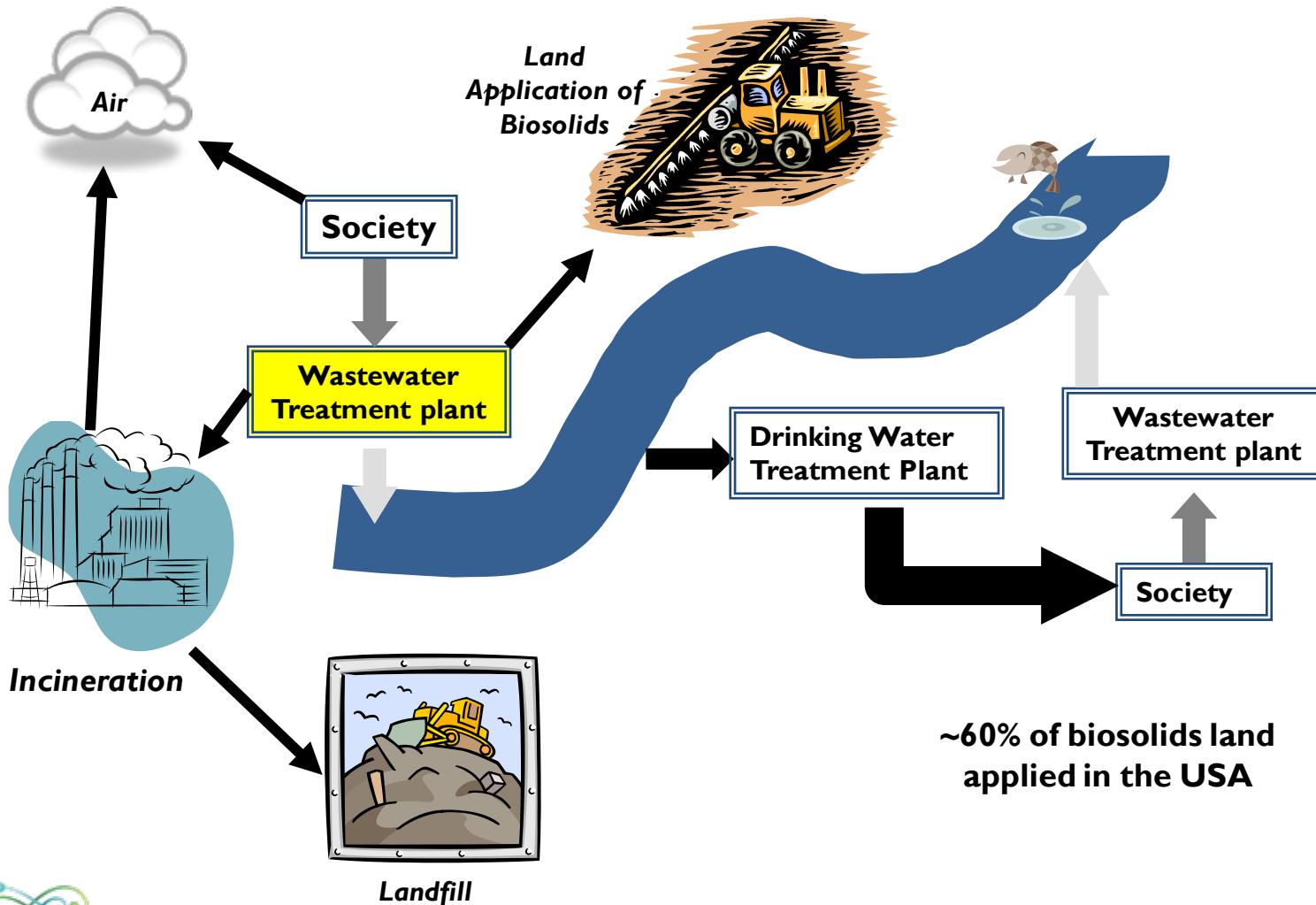




Fun – Complex Stuff

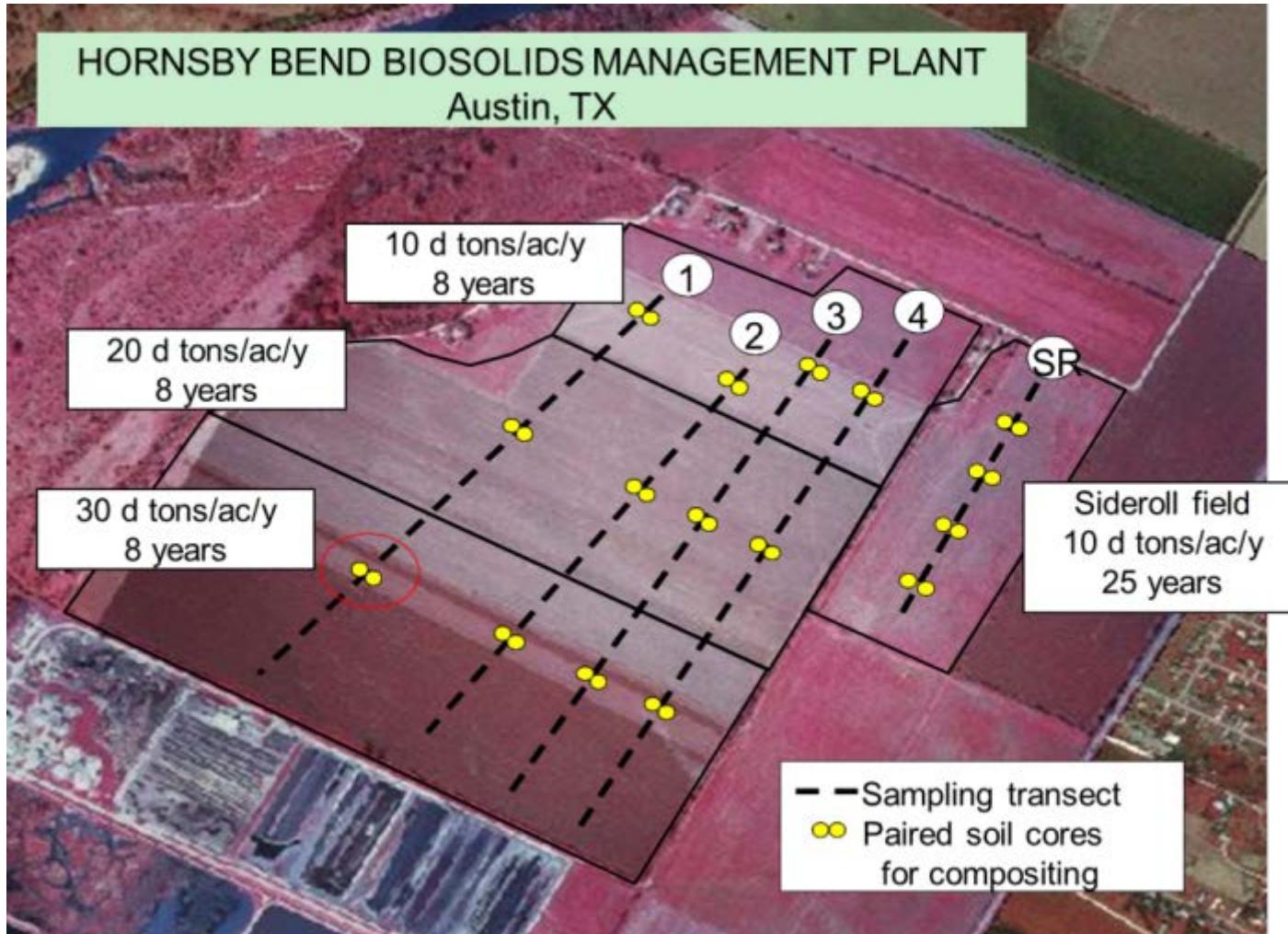


Where do Biosolids Go?



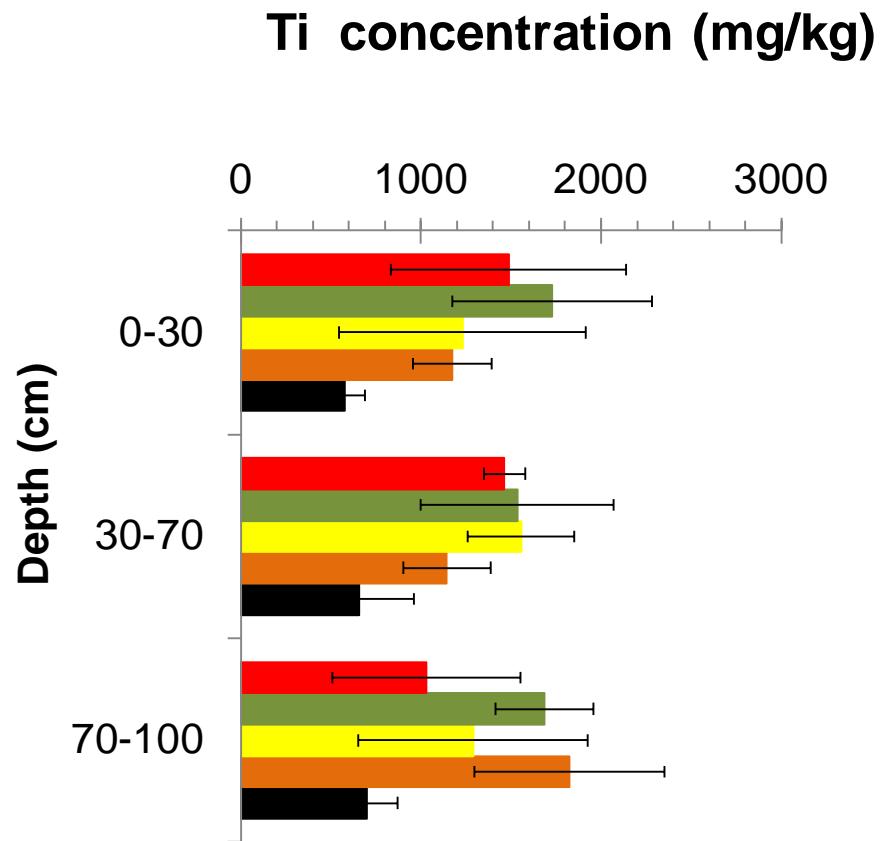
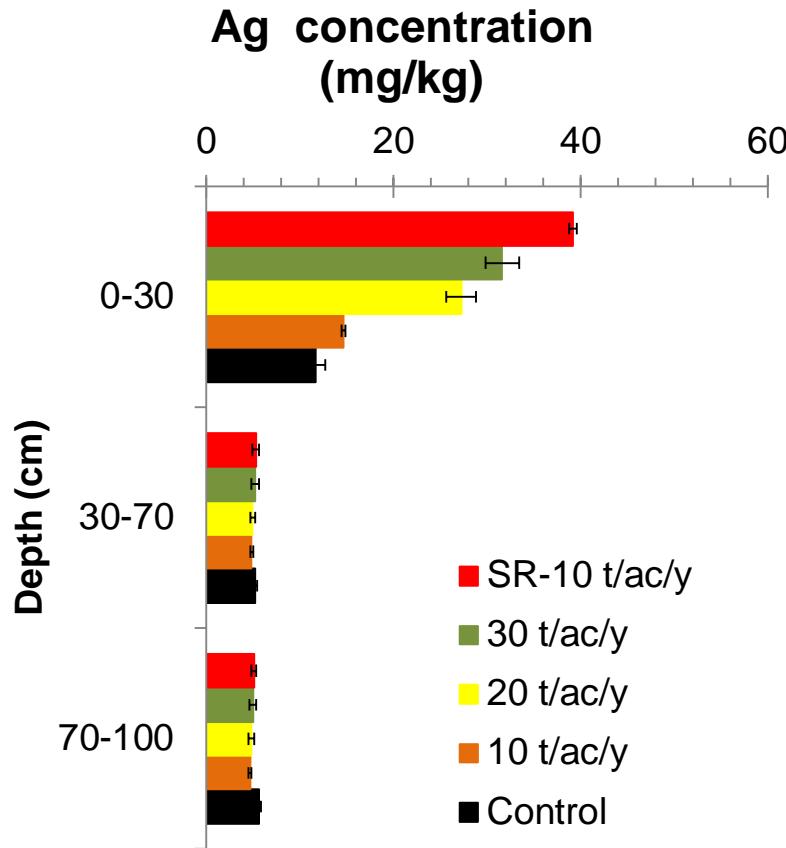


USDA Field Site Near Austin, TX Long-term Biosolid Applications



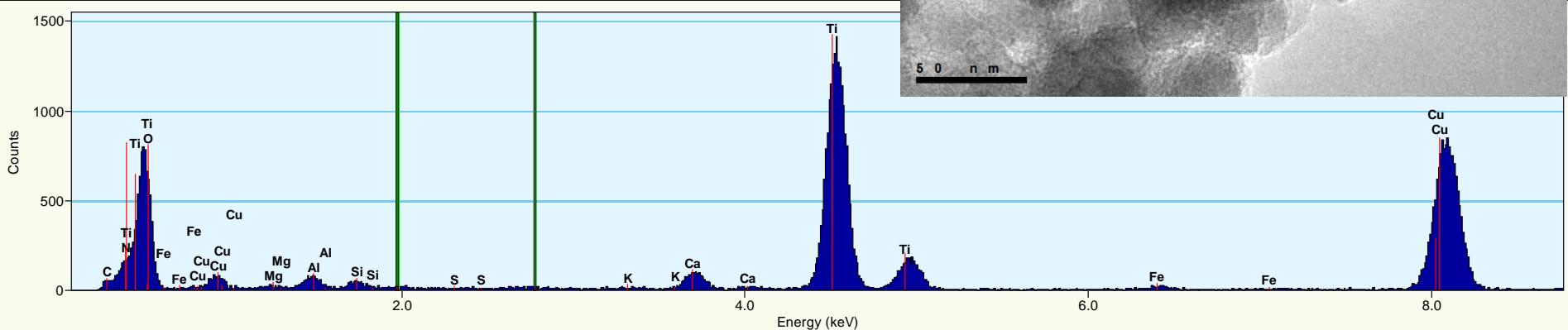
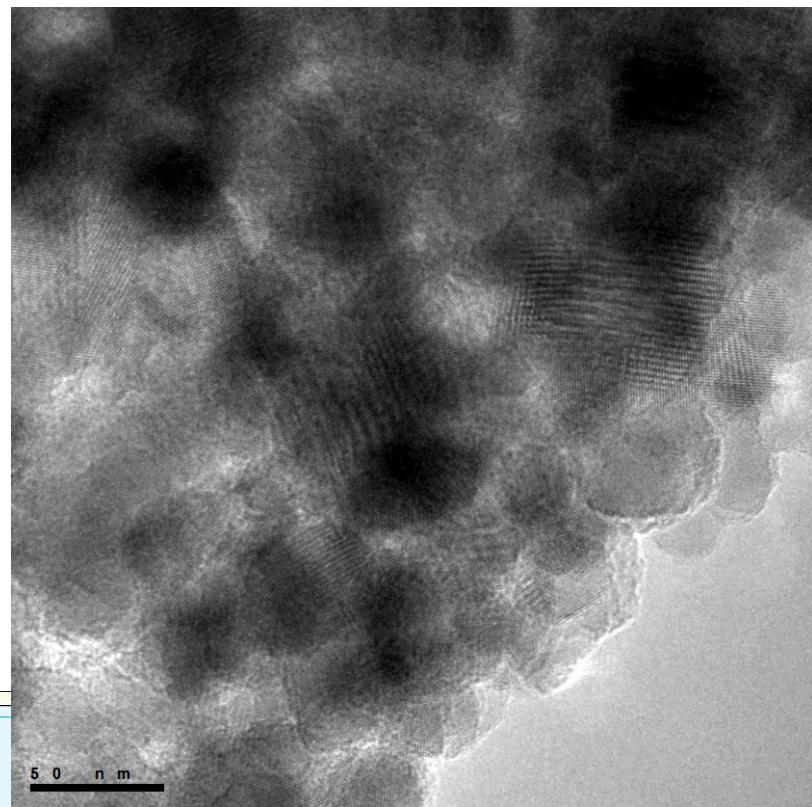
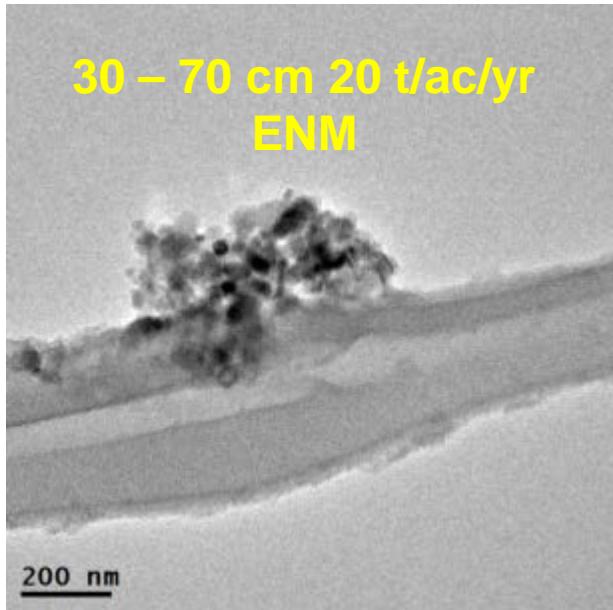


Each element has different soil profile after applying biosolids





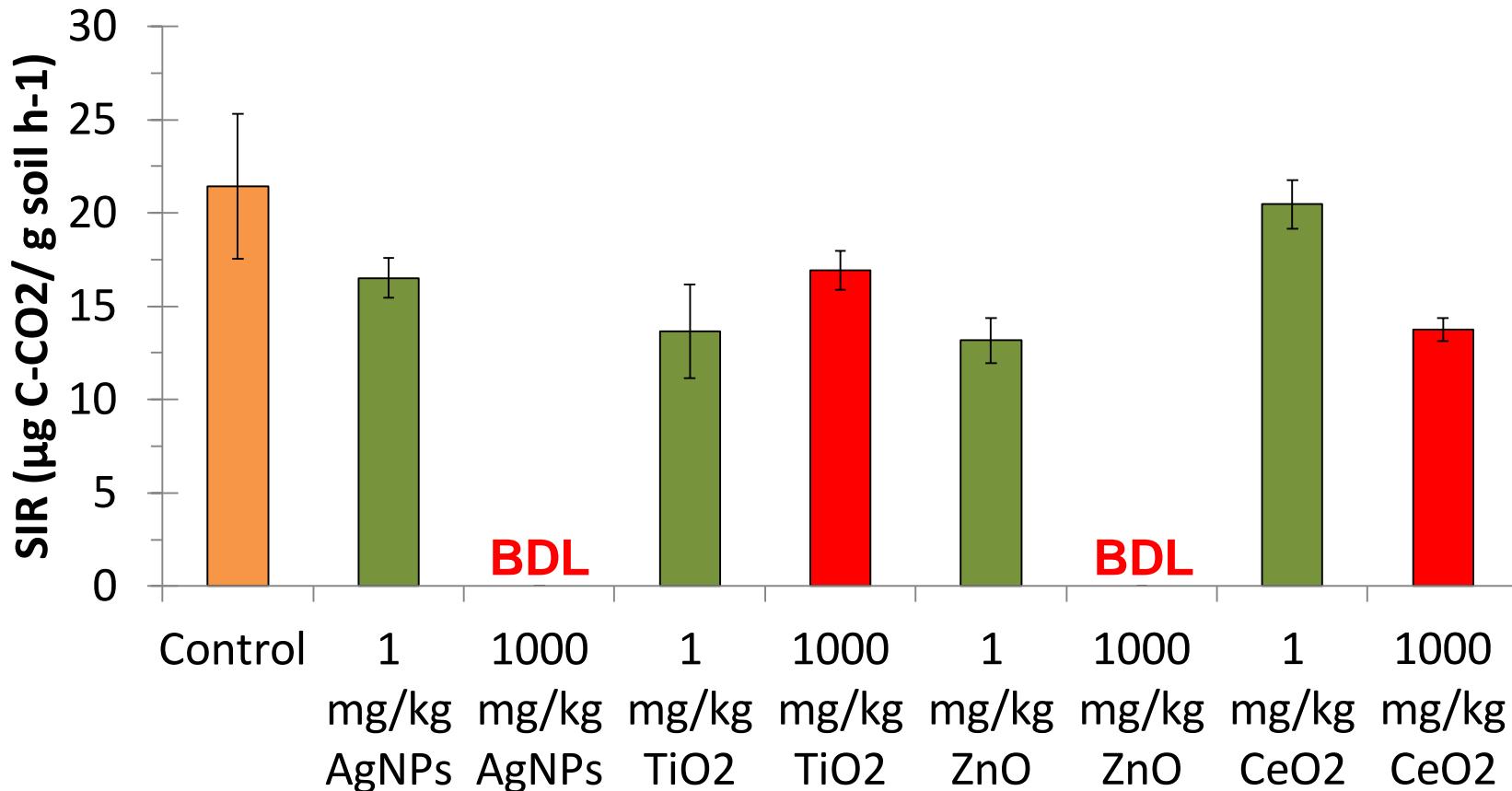
TiO₂ Nanomaterials are present in Biosolid Ammended Fields





Do NM_s in Biosolids Amended Soils Impact Biological Activity ?

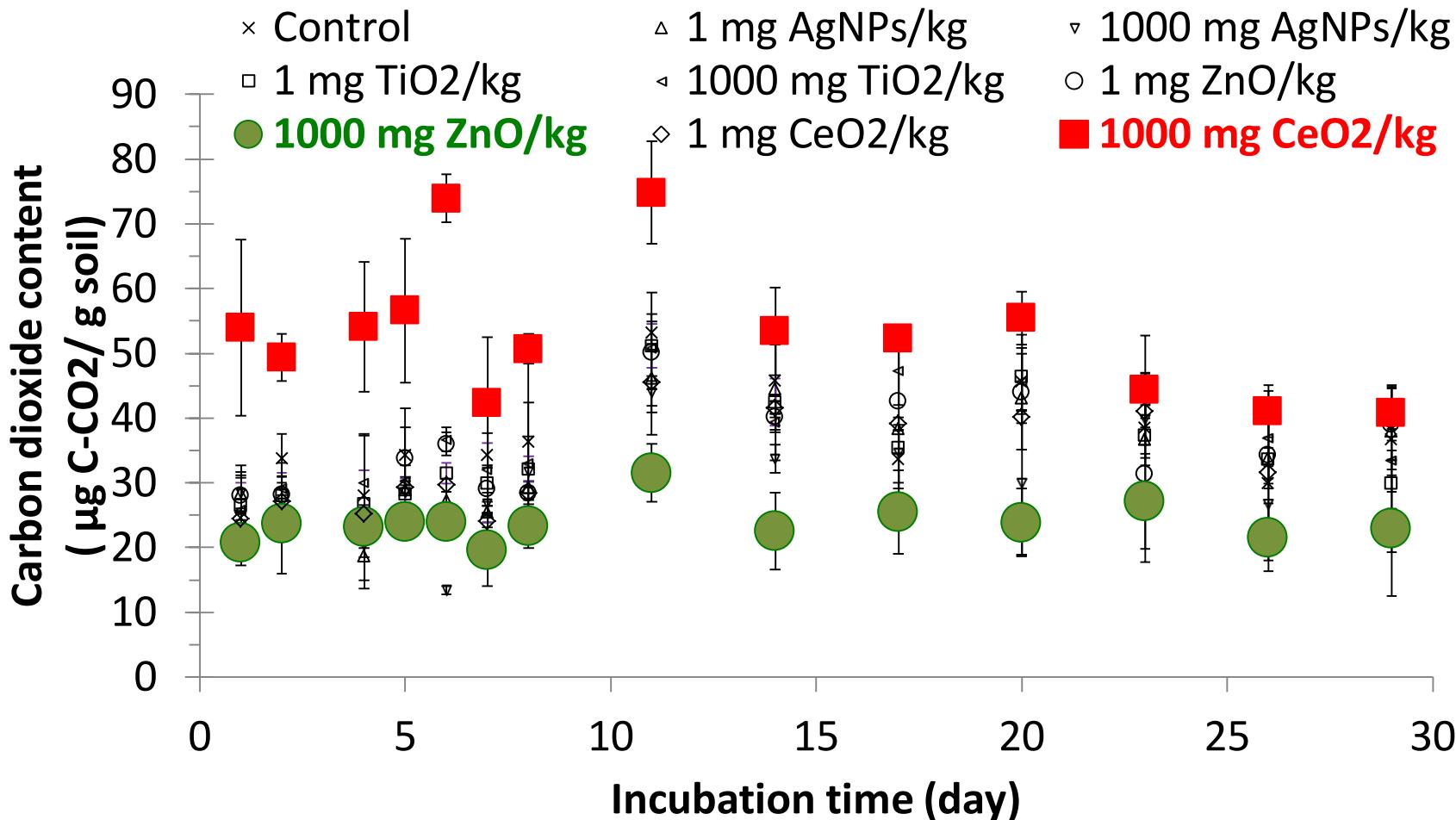
Substrate Induced Respiration (28 day)





Basal Respiration Tests

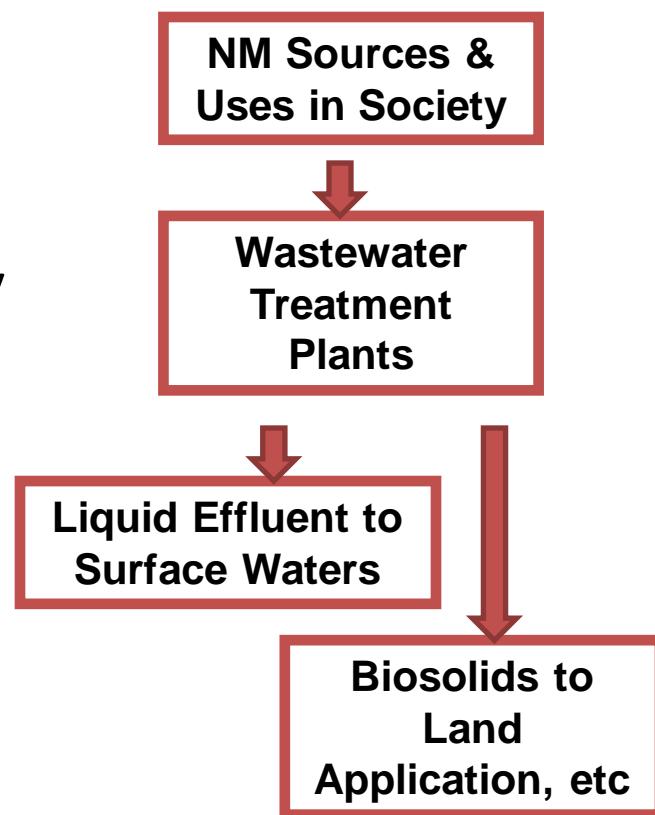
CeO₂ increases CO₂ production





Summary

- Nanomaterials will be emitted from various industrial, commercial and residential products into sewers
- On-site treatment systems (lime, membranes) can be effective
- Uniqueness of Nanomaterials: They “accumulate” at **interfaces**
- At current levels, engineered NMs are UNLIKELY to impact WWTP operations





Acknowledgements

- Water Environment Research Foundation (WERF)
 - Paul L. Busch Award 2006
 - Project U1R10
- NIH/NIEHS - NIH/NIEHS Nano-Go Funding: RES018801Z
- NSF (*CBET 1336542*)
- EPA (RD83322 & RD835580)
- SRC 425.040 & 425.052
- Contact: p.westerhoff@asu.edu



DOE BER
Grant number: DE-FG02-08ER64613





Upcoming Webinars

Monday, June 13, 2016 - 1pm Eastern Time

Self-Powered Wearable Devices for Monitoring
Personal Health

Presenter: Veena Misra, Ph.D.

Director, NSF ASSIST Nanosystems Center

Professor, Department of Electrical Engineering



The effects of slurries on the proliferation, viability, or membrane integrity of model organisms.



Table 4 The effects of slurries on the proliferation, viability, or membrane integrity of model organisms. IC-50 levels were not determined (ND) under several conditions as noted

Assay	IC-50 (mg mL ⁻¹)			
	c-SiO ₂	f-SiO ₂	CeO ₂	Al ₂ O ₃
Bioluminescence of <i>A. fischeri</i>	ND ^a	ND ^b	ND ^c	ND ^d
Proliferation of A549 cells	3.8 ± 1.3	3.6 ± 0.2	ND ^e	ND ^f
Viability of A549 cells	1.2 ± 0.2	1.5 ± 0.2	ND ^g	ND ^h
Integrity of A549 cells	4.6 ± 0.2	3.1 ± 0.2	ND ^g	ND ^h



Fig. 11 Ecotoxicity findings determined in bioassays with freshwater crustaceans (*Cladocera*) (A) and algae (B) for three major classes of metal-based nanoparticles used in CMP slurries. Note: *D. similis* = *Daphnia similis*; *D. magna* = *Daphnia magna*; *D. pulex* = *Daphnia pulex*; *C. dubia* = *Cerodaphnia dubia*; *P. subcapitata* = *Pseudokirchneriella subcapitata*. Legends: Data for CeO₂ (■), Al₂O₃ (□), and SiO₂ (▨).

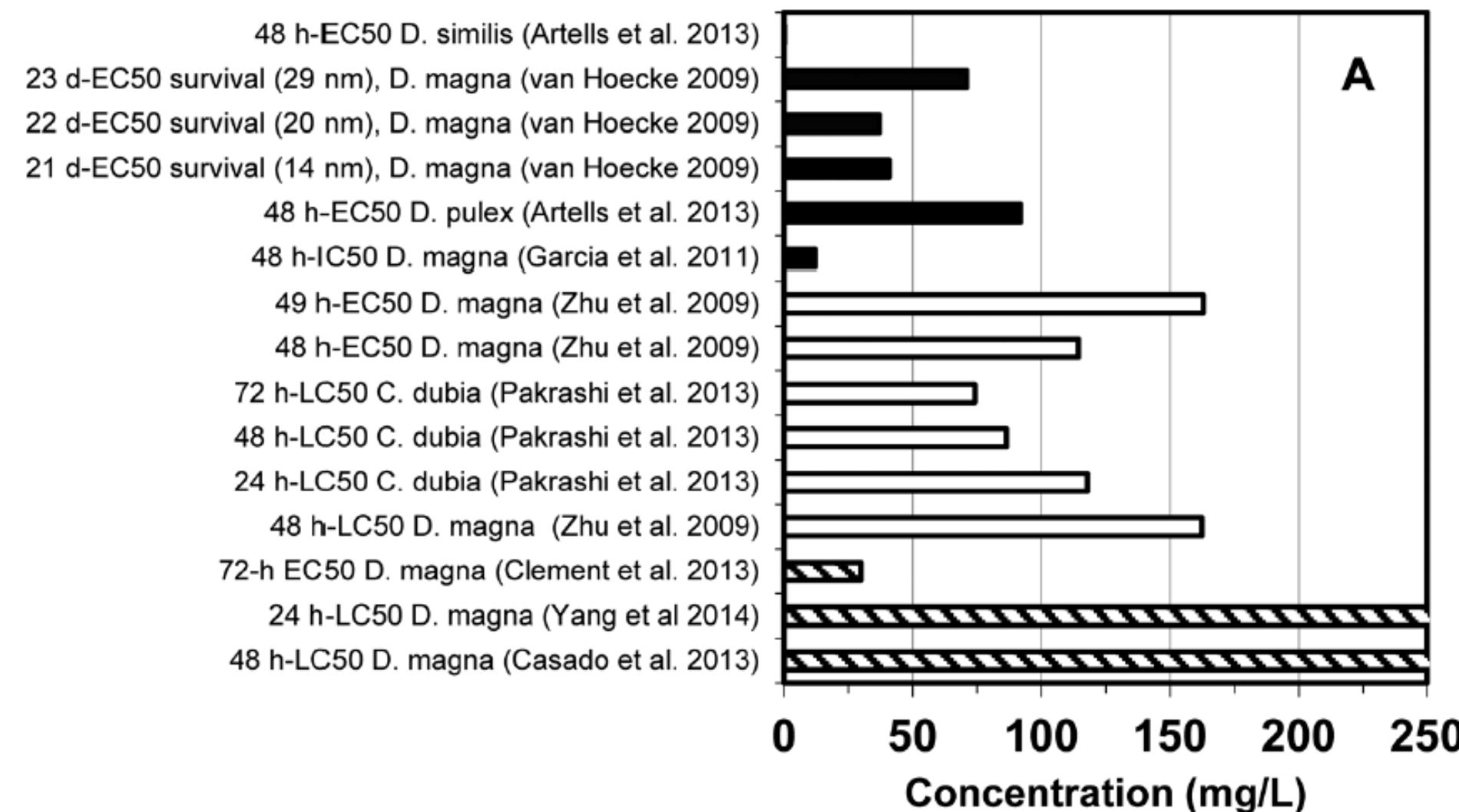
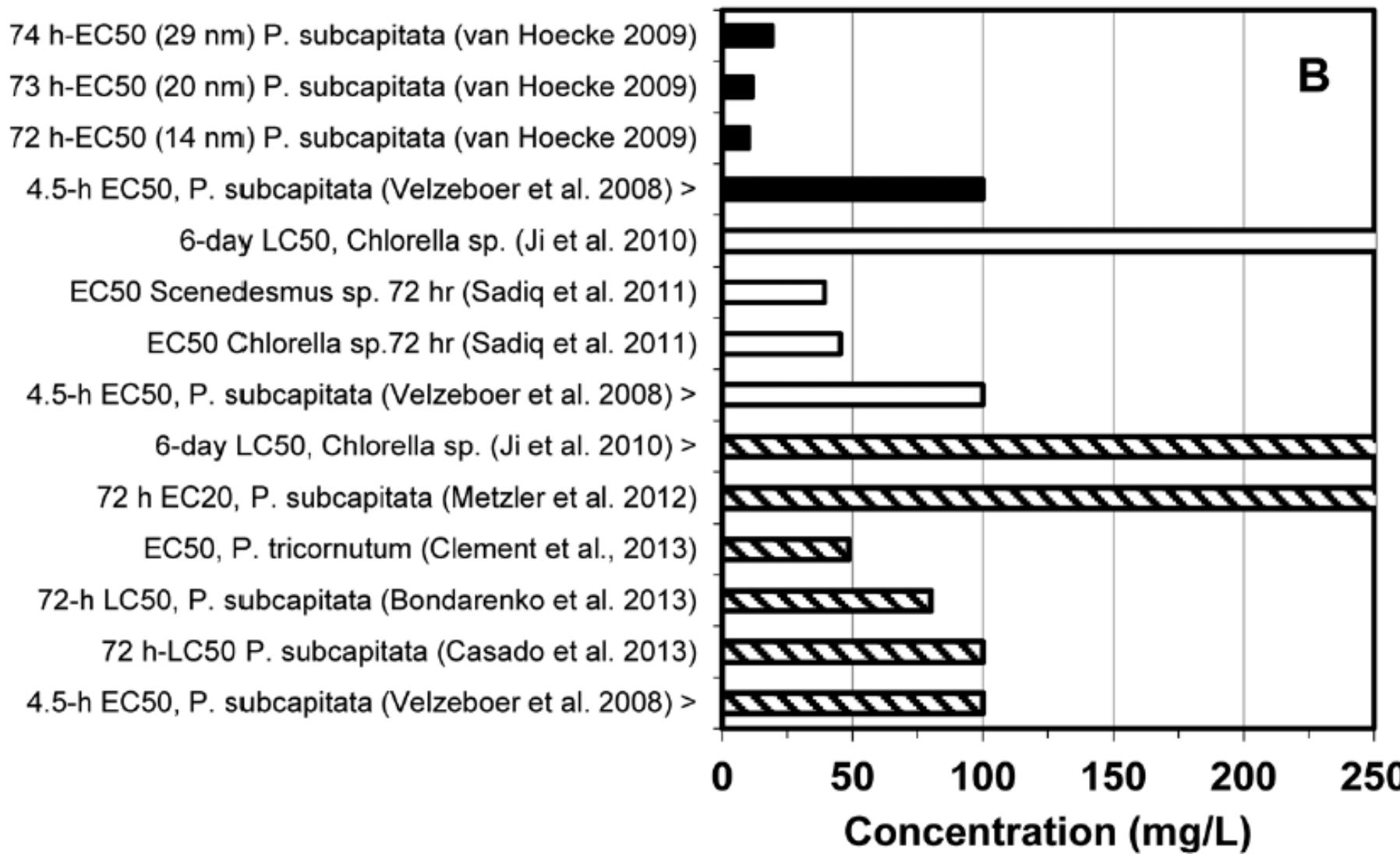




Fig. 11 Ecotoxicity findings determined in bioassays with freshwater crustaceans (*Cladocera*) (A) and algae (B) for three major classes of metal-based nanoparticles used in CMP slurries. Note: *D. similis* = *Daphnia similis*; *D. magna* = *Daphnia magna*; *D. pulex* = *Daphnia pulex*; *C. dubia* = *Cerodaphnia dubia*; *P. subcapitata* = *Pseudokirchneriella subcapitata*. Legends: Data for CeO₂ (■), Al₂O₃ (□), and SiO₂ (▨).





Lifecycle Perspective of NMs

Societal Use of
Products

Measuring Releases of
NPs in various media



NPs in biosolids that are
land applied

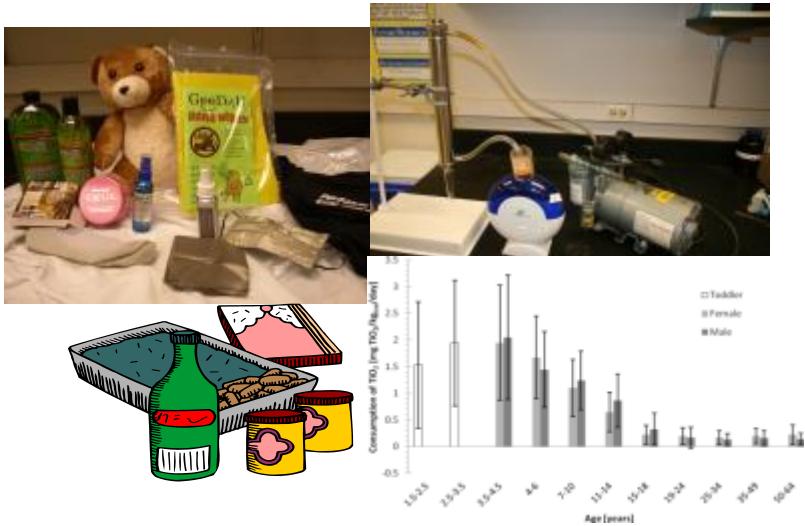
Removal of NPs at WWTPs





Lifecycle Perspective of NMs

Societal Use of Products



NPs in biosolids that are land applied

Measuring Releases of NPs in various media

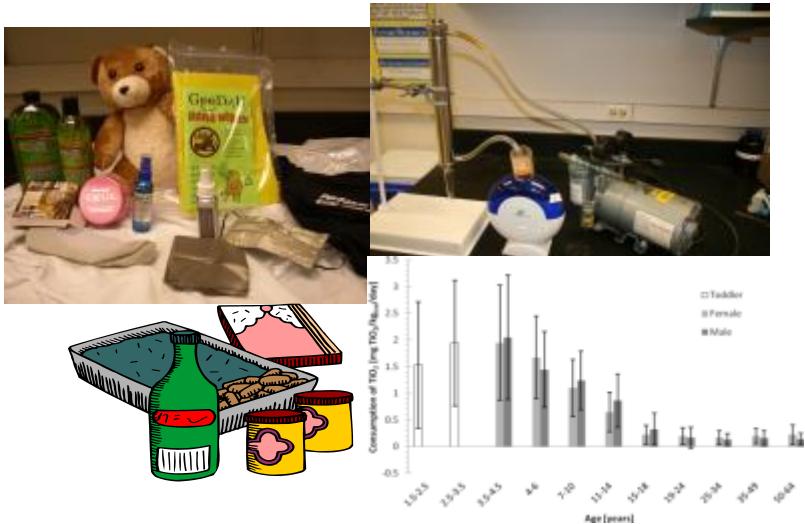


Removal of NPs at WWTPs

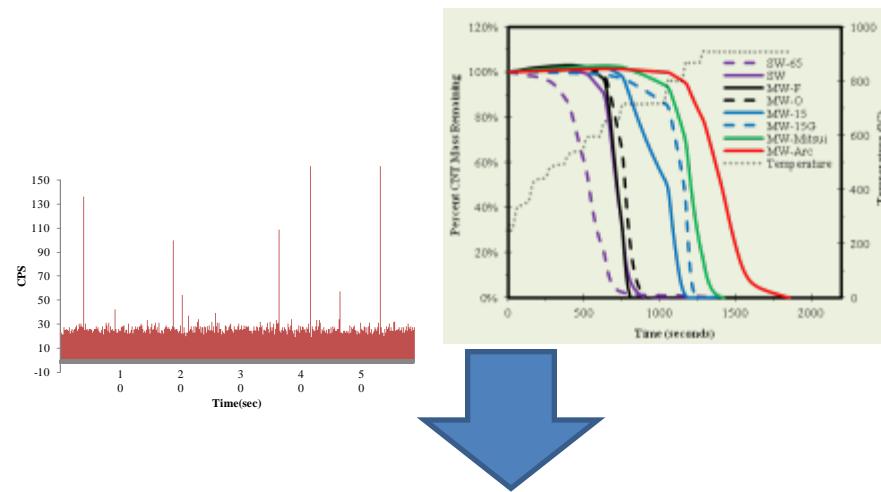


Lifecycle Perspective of NMs

Societal Use of Products



Measuring Releases of NPs in various media

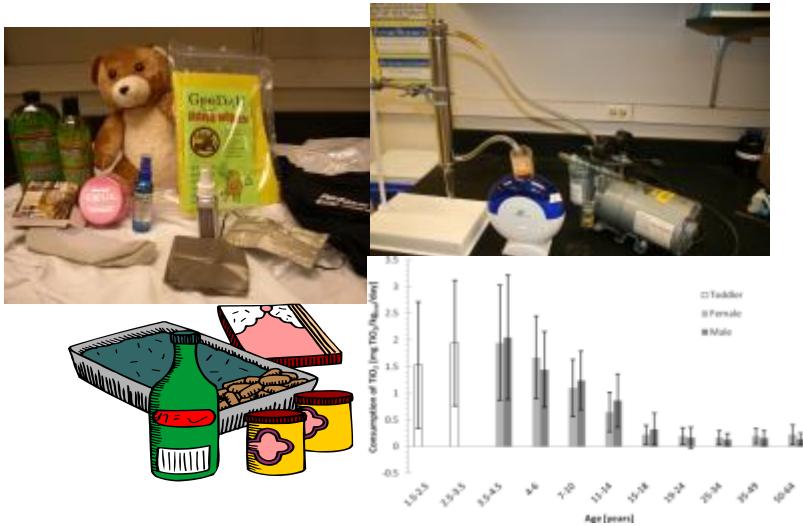


NPs in biosolids that are land applied

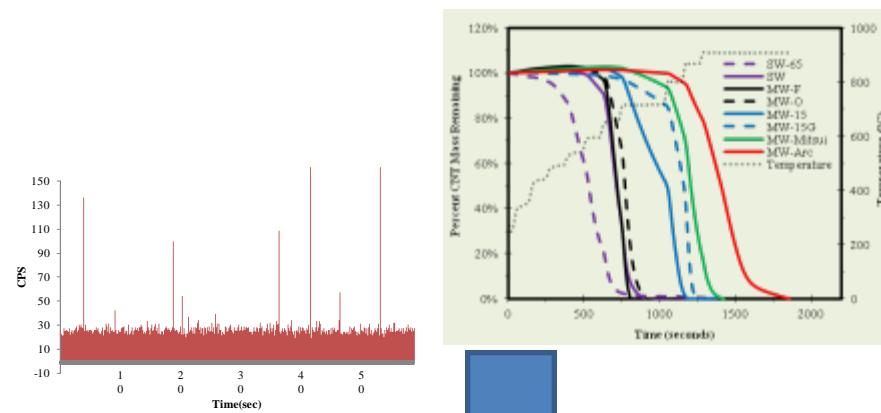
Removal of NPs at WWTPs

Lifecycle Perspective of NMs

Societal Use of Products



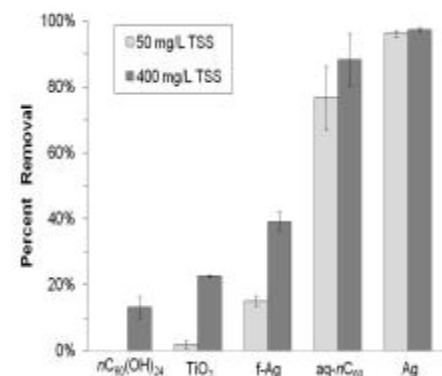
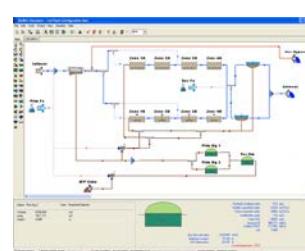
Measuring Releases of NPs in various media



NPs in biosolids that are land applied

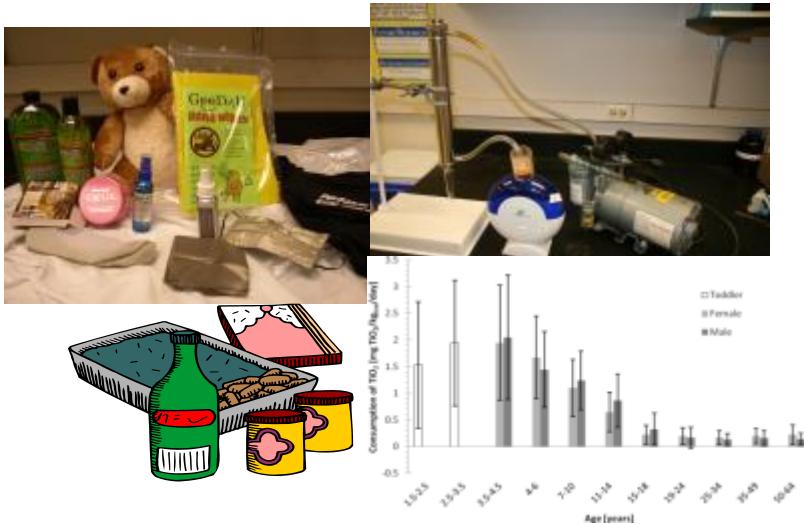


Removal of NPs at WWTPs

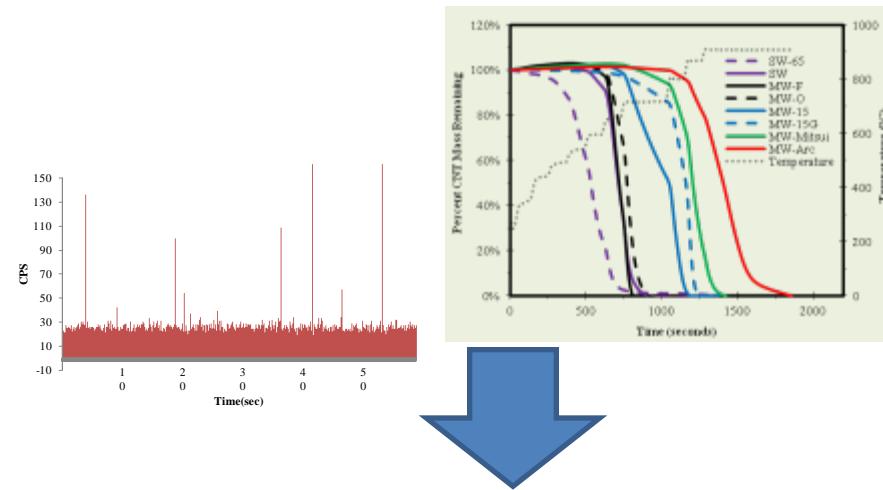


Lifecycle Perspective of NMs

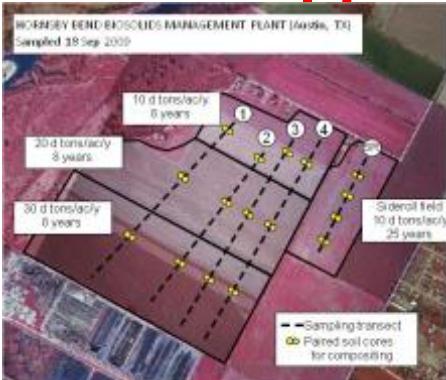
Societal Use of Products



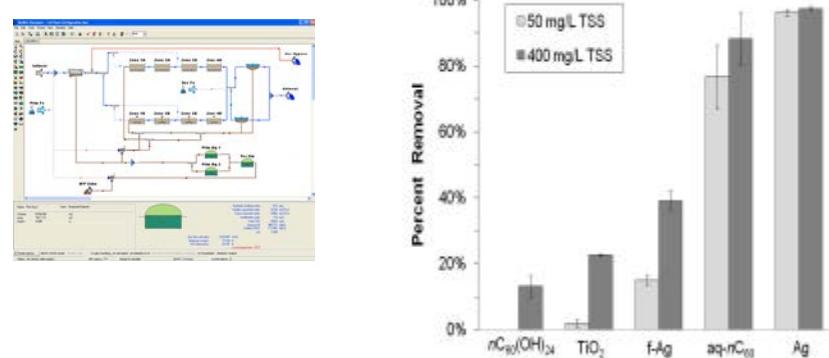
Measuring Releases of NPs in various media



NPs in biosolids that are land applied



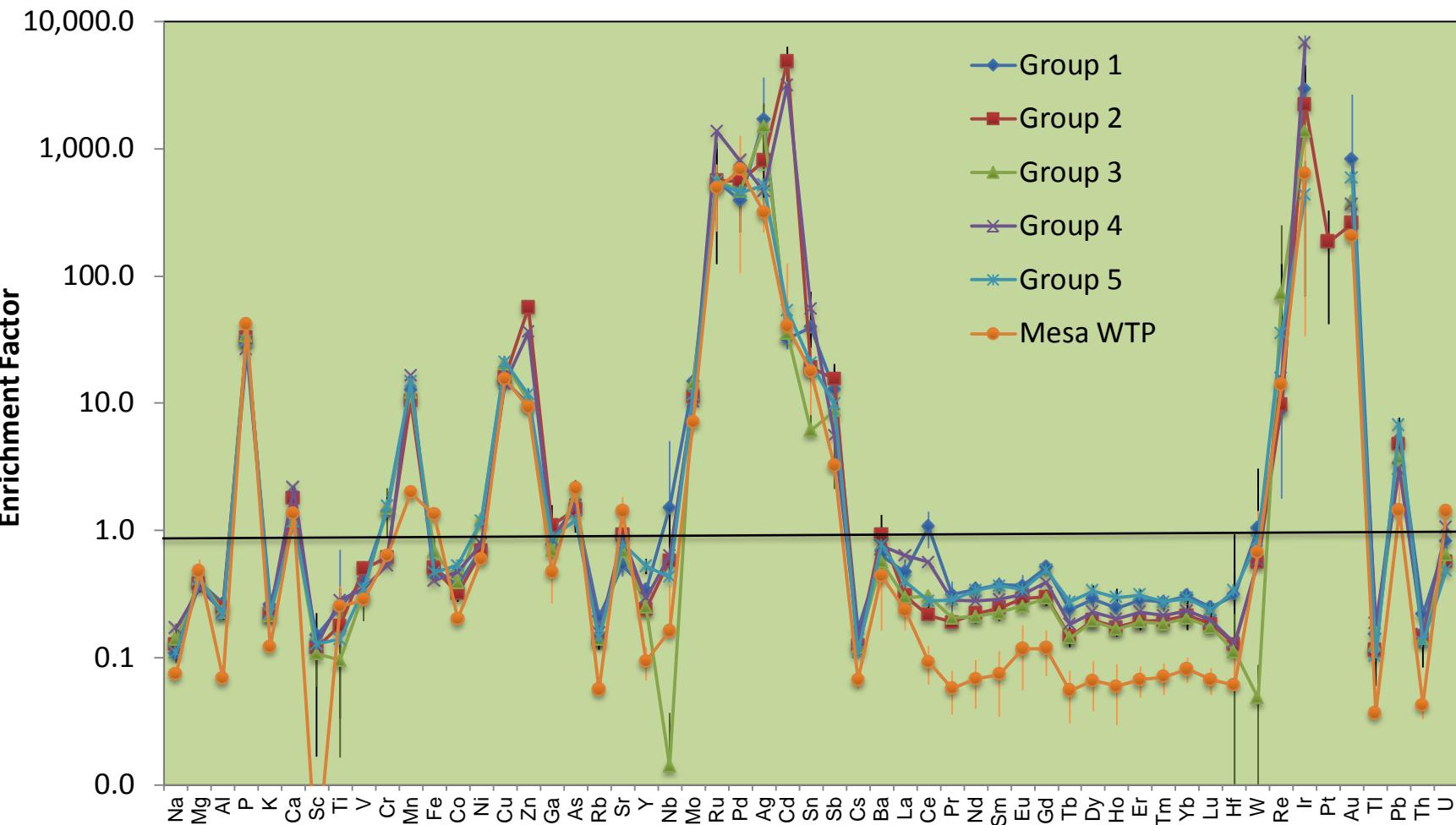
Removal of NPs at WWTPs





Big Picture – EPA Composite Biosolids

Biosolid concentrations, normalized to Upper Continental Crust



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Thank You!

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